

Ammonia (see worksheet) \_\_\_\_\_ without reacting with \_\_\_\_\_

1. \_\_\_\_\_ colorless, pungent \_\_\_\_\_ color \_\_\_\_\_ odor \_\_\_\_\_

2. \_\_\_\_\_ colorless, odorless \_\_\_\_\_ color \_\_\_\_\_ odor \_\_\_\_\_

3. \_\_\_\_\_ very \_\_\_\_\_ compounds that release volatile \_\_\_\_\_ from \_\_\_\_\_

Procedure: Add 10 mL "A" (see text), "B" (see text), and "C" (see text).

\_\_\_\_\_

| Substance                              | Acid (pH) / Base (pH) or Salt (pH) | Reaction with Mg (M or MgO) | Conductivity (C or MC) |
|----------------------------------------|------------------------------------|-----------------------------|------------------------|
| $\text{HNO}_3$                         | _____                              | _____                       | _____                  |
| $\text{NaOH}$                          | _____                              | _____                       | _____                  |
| $\text{H}_2\text{SO}_4$                | _____                              | _____                       | _____                  |
| $\text{Na}_2\text{SO}_4$               | _____                              | _____                       | _____                  |
| $\text{HCl}$                           | _____                              | _____                       | _____                  |
| $\text{LiOH} \cdot \text{H}_2\text{O}$ | _____                              | _____                       | _____                  |
| $\text{KNO}_3$                         | _____                              | _____                       | _____                  |
| $\text{NaCl}$                          | _____                              | _____                       | _____                  |
| $\text{HCl} \cdot \text{H}_2\text{O}$  | _____                              | _____                       | _____                  |

# Note Taking Guide 1101 Key

**Ying Liu**



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