Physical Science A Final Exam Study Guide

- Martter
 - Compare and contrast physical and chemical properties, and physical and chemical changes
 - Compare and contrast properties and behaviors of atoms in each of the 3 states of matter
 - Convert between mass, volume, and density, and relate appropriate SI units for each variable
 - Predict how solutions of different densities would settle in a graduated cylinder
- · Atoms, Elements, and the Periodic Table
 - Describe Antoine Lavoisier and Dmitiri Mendeleev's contributions to the development of the modern periodic table
 - Define the term ion and be able to predict the most common ions formed for elements by looking at the periodic table
 - Identify three subatomic particles and compare their properties
 - Compare and contrast atoms in ground and excited states
 - Distinguish the atomic number of an element from the mass number of an isotope, and use these numbers to describe the structure of atoms
 - Describe the currently accepted model of an atom

numbers to describe the attaches of assess

Company and contrast sens and constant bonds.

Provided this periodical to all charminal manufactures.

Exprison the action rule for bonding

CONTRACTOR STREET

properties, plants train and unbrief observed

Explain the rate that calenda electrons play in bonding

Mentify three substants particles and compare their properties.
 Compare and contrast above in ground and excited states.

Desput the Sections that can be used to recrease reaction when

Balance Carried Specifics in Carried by Carried Carried

Apply the Law of Comprisation of Mass to chemical reach
 List was to tell that a chemical market has accurred

Distinguish Seriason analytication and particular to tag from

- Interpret the information provided about the atoms and elements in a balanced chemical reaction
- Differentiate between elements that are good and poor conductors of heat and electricity

Product the reactivity of some electronic based on their bacations in the periods table

coll the represents that make up polymers for different bodygical macromolecules

Classify charming married as synthesis, decomposition, single-replacement, dis-

Interpret channel of equations in factor of machines, products, and consequation of mac-

- Understand the development and organization of the periodic table, especially in terms of group properties, atomic mass and valence electrons
- Identify general properties of metals, nonmetals, and metalloids

Company and contract Baths and Ruther ford's attempt models.

- Define the words malleable, brittle, volatile, and ductile
- Explain how the electron cloud model represents the behavior and locations of electrons in atoms

Distinguish the plants number of an element from the trace number of an instage, and use flows

Understand the development and organization of the particle table, expectable is formed a proper

Draw Lewis Det diagrams for elements, and use them to predict borning between one or more atomic

Calculate products of reactions or products by using malar mass, mole ratios, and balanced charms at

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