

Chapter 12 – Stoichiometry

Note Taking Guide: Episode 801 Stoichiometry

-Study of the **QUANTITY** relationships in a **CHEMICAL REACTION**

-Based on **BALANCED** equations



1. Particles

2 atoms Mg

2 atoms O

2 atoms Mg

2 atoms O

2 atoms Mg

1 molecule O₂

2 molecule MgO

10 atoms Mg

5 molecule O₂

10 molecule MgO

2 (6.02 x 10²³)
atoms Mg

1 (6.02 x 10²³)
molecule O₂

2 (6.02 x 10²³)
molecule MgO

2. Moles

2 mol Mg

1 mol O₂

2 mol MgO

The **COEFFICIENTS** in a **BALANCED EQUATION** give the **MOLE RATIO** for the **SUBSTANCES** involved in the **REACTION**.

Lab: Episode 801 - Can You Find the Mole Ratio?

In this experiment, baking soda (NaHCO₃) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two.

Data Table:

mass of test tube and baking soda	19.46 g
mass of empty test tube	19.21 g

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A Loxley



Note Taking Guide Episode 801 Stoichiometry:

Whispering the Secrets of Language: An Mental Journey through **Note Taking Guide Episode 801 Stoichiometry**

In a digitally-driven earth wherever displays reign great and immediate interaction drowns out the subtleties of language, the profound techniques and emotional subtleties hidden within phrases frequently move unheard. Yet, situated within the pages of **Note Taking Guide Episode 801 Stoichiometry** a fascinating fictional value pulsating with fresh emotions, lies an extraordinary journey waiting to be undertaken. Written by a talented wordsmith, this wonderful opus attracts viewers on an introspective journey, gently unraveling the veiled truths and profound influence resonating within the very material of every word. Within the mental depths of the moving review, we shall embark upon a honest exploration of the book is core themes, dissect its fascinating publishing type, and yield to the strong resonance it evokes heavy within the recesses of readers hearts.

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