

REPORT FOR EXPERIMENT 9

Single Displacement Reactions

NOTE: In this experiment, copper ions will have 2+ charge.

NAME Tate
LAB PARTNER(S) _____
DATE 11/1/20

Evidence of Reaction

Describe any evidence of reaction; if no reaction was observed, write "None".

Equation (to be completed)
Write "No reaction", if no reaction was observed.

1. Changed color decomposing



2. change in color sparkly



3. none



4. none



5. bubbling/dissolving



6. none



QUESTIONS AND PROBLEMS

1. Complete the following table by writing the symbols of the two elements whose reactivities are being compared in each test:

| | Tube Number | | | | | |
|------------------|-------------|-----------|-----------|-----------|-----------|-----------|
| | 1 | 2 | 3 | 4 | 5 | 6 |
| Greater activity | <u>Cu</u> | <u>Zn</u> | <u>Mg</u> | <u>H</u> | <u>Zn</u> | <u>Pb</u> |
| Lesser activity | <u>Ag</u> | <u>Pb</u> | <u>Zn</u> | <u>Cu</u> | <u>H</u> | <u>Cu</u> |

2. Arrange Pb, Mg, and Zn in order of their activities, listing the most active first.

(1) Mg

(2) Zn

(3) Pb

Report For Experiment 1single Displacement Reactions

JG Myers



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