

Answer following



- 1. Substrate and temperature: 20°C 30°C 40°C
- 2. A graph of initial velocity: $\text{mg ml}^{-1} \text{min}^{-1}$ vs. mg ml^{-1} substrate concentration
- 3. A graph of initial velocity: $\text{mg ml}^{-1} \text{min}^{-1}$ vs. mg ml^{-1} substrate concentration

Calculate all values of K_m and V_{max} from all graphs in the following figure. Please show the steps in your calculations.

- 1. Lineweaver-Burk plot: $1/v$ vs. $1/S$
- 2. Eadie-Hofstee plot: v vs. v/S
- 3. Hanes-Woolf plot: v/S vs. v
- 4. Michaelis-Menten plot: v vs. S



Answer 1-4

Graph 1-4 show the initial velocity of the reaction at different substrate concentrations.



Graph 5-8 show the initial velocity of the reaction at different substrate concentrations.



Graph 9-12 show the initial velocity of the reaction at different substrate concentrations.

Prentice Hall Geometry 4 5 Answers

S Marginson



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