

Chapter 10 - Chemical Quantities

10.1 The Mole: A Measurement of Matter:

Mole: (Thank, Google's, videos)

- Unit of measurement

Representative Particles:

- Smallest unit of a substance (atom, molecule, formula unit, ions, anything...)
- Avogadro's number = 6.02×10^{23} representative particles (RP)
- 6.02×10^{23} representative particles = 1 mol

Molar Mass:

- Molar mass = g/mol
- Molar mass of:
 - an element = atomic mass (g)
Ex. 24.3 g/mol = molar mass of Mg
 - a compound = sum of the atomic masses (g) of each element
Ex. 58.44g/mol = molar mass of NaCl
- Molar mass = 1 mol

Molar Volume:

- Molar volume = 22.4 L/mol
- Standard temperature and pressure (STP) =
 - temp = 0°C
 - pressure = 1 atm
- 22.4 L = 1 mol (at STP)

Calculating Molar mass from Density:

- molar mass (g/mole) = density at STP (g/L) $\times$ molar volume at STP (22.4L/mole)

$$\text{Molar mass (g/mol)} = \left(\frac{\text{grams}}{\text{L}} \right) \left(\frac{22.4 \text{ L}}{\text{mole}} \right)$$

Calculating Density from Molar mass:

- Density at STP (g/L) = molar mass (g/mole) $\times$ molar volume at STP (1mole/22.4L)

$$\text{Density (g/L)} = \left(\frac{\text{grams}}{1 \text{ mole}} \right) \left(\frac{1 \text{ mole}}{22.4 \text{ L}} \right)$$

Prentice Hall Chemistry Chemical Quantities Chapter 10

Lecture Notes

Guillaume Favre



Prentice Hall Chemistry Chemical Quantities Chapter 10 Lecture Notes:

Whispering the Techniques of Language: An Emotional Journey through **Prentice Hall Chemistry Chemical Quantities Chapter 10 Lecture Notes**

In a digitally-driven world wherever displays reign great and quick connection drowns out the subtleties of language, the profound secrets and mental nuances concealed within words frequently get unheard. However, situated within the pages of **Prentice Hall Chemistry Chemical Quantities Chapter 10 Lecture Notes** a charming fictional treasure sporting with organic emotions, lies an extraordinary journey waiting to be undertaken. Published by a talented wordsmith, this charming opus encourages readers on an introspective journey, gently unraveling the veiled truths and profound influence resonating within the very material of each word. Within the psychological depths with this emotional evaluation, we shall embark upon a heartfelt exploration of the book is key styles, dissect their captivating publishing design, and fail to the effective resonance it evokes heavy within the recesses of readers hearts.

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