#### Macromolecules - Answers

- An amino acid contains a central <u>carbon</u> atom, a hydrogen atom, a carboxyl group, an <u>amino group</u> group, and a differing group.

  R group.
- What is a dipeptide? What type of bond is involved? = two amino acids joined via a peptide bond.
- State two functions of fats. long term energy storage insulation lubrication cushioning
- What is the structure of a phospholipid? hydrophilic head group, with a phosphate always charged and a group of atoms (ex. choline, serine); two hydrophobic tails, fatty acids with long hydrocarbon chains
- 6. ATP is a molecule that stores a great deal of energy within its phosphate bonds. True or False?
  True
- Name and state the structure of a common dissaccharide.
   two sugars joined via covalent bond; ex. glucose-glucose = maltose; glucose-fructose = sucrose
- 8. Name and state the structure of a common polysaccharide. glycogen and starch = polymers of alpha glucose, storage form of glucose in animals and plants cellulose = polymers of beta glucose, rigid structure in plant cell walls chitin = polymer of glucosamine, very strong and flexible, found in exoskeletons of arthropods
- Heating an enzyme to a certain temperature usually destroys its enzymatic activity even though
  the same sequence of amino acids remains. Name and explain the levels of structure that
  have been affected by heating.

secondary structure = alpha belices, beta pleated sheets, random coils tertiary structure = overall three dimensional conformation, or shape, of the polypeptide quaternary structure = assembly of subunits

primary structure = chain of amino acids; usually remains intact upon denaturation due to strong peptide bonds.

# **Prentice Hall Biology Study Guide Macromolecules Answers**

Jianjun Gao

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