

Name: _____ Date: _____

CHEMISTRY CHEMICAL REACTIONS LAB #2

Go to the website https://javalab.org/en/precipitation_reaction_en/

- You will mix dissolved ionic compounds together with an expected double replacement reaction. Only some of the mixtures will react. Some of the mixtures will not react.
- Identify the product ionic compound that forms the insoluble precipitate (the solid that flocks out of solution).
- Identify the spectator ions—the ions that remain dissolved and inert after the other compound precipitates.

Instructions

1. Follow the recipe. Add the compounds and the correct number of molecules to the beaker. The number in parentheses () is the number of molecules to add.
2. Watch the mixture for about 1-2 minutes. The ions (cations and anions) will disperse in the water and move randomly. Enough time must pass for the ions to bump repeatedly into other ions.
3. Column 1: If NO REACTION happens, choose "NO". If a chemical reaction happens, choose "YES".
4. Column 2: If NO REACTION happens, choose "NONE". If a chemical reaction happens, choose the solid product (the precipitate) formed during the reaction. A precipitate is a solid ionic compound that forms during a double replacement reaction.
5. Column 3: If NO REACTION happens, choose "NONE". If a chemical reaction happens, choose the spectator ions that remain dissolved in water.
6. Use RESET button to clear the beaker. Add the next recipe.

REACTANTS		Reaction	Precipitate Chem Formula	Spectator Ions
(6) Na ₂ S	(6) BaCl ₂			
(6) Na ₂ S	(6) Ca(NO ₃) ₂			
(6) Na ₂ S	(6) KI			
(6) Na ₂ S	(6) AgNO ₃			
(6) Na ₂ S	(6) Pb(NO ₃) ₂			
(6) BaCl ₂	(6) KI			

Precipitation Reactions Answer Sheet

JL Elias



Precipitation Reactions Answer Sheet:

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Quantitative Analysis Laboratory Manual Dr. Bupinderpal Singh , Dr. Bikramjit Singh, Mr. Akash Bans, 2025-06-29

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predicted that the global supply of rare earth elements could soon be exhausted A sustainable approach to the use and recovery of rare earth elements is needed and this book addresses the political economic and research agendas concerning them The problem is discussed thoroughly and a multi disciplinary team of authors from the chemistry engineering and biotechnology sectors presents a range of solutions from traditional metallurgical methods to innovations in biotechnology Case studies add value to the theory presented and indirect targets for recovery such as municipal waste and combustion ash are considered This book will be essential reading for researchers in academia and industry tackling sustainable element recovery as well as postgraduate students in chemistry engineering and biotechnology Environmental scientists and policy makers will also benefit from reading about potential benefits of recovery from waste streams AP Chemistry Premium, 2025: Prep Book with 6 Practice Tests + Comprehensive Review + Online Practice Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan, 2024-07-02 Be prepared for exam day with Barron's Trusted content from AP experts Barron's AP Chemistry Premium 2025 includes in depth content review and practice It's the only book you'll need to be prepared for exam day Written by Experienced Educators Learn from Barron's all content is written and reviewed by AP experts Build your understanding with comprehensive review tailored to the most recent exam Get a leg up with tips strategies and study advice for exam day it's like having a trusted tutor by your side Be Confident on Exam Day Sharpen your test taking skills with 6 full length practice tests 3 in the book and 3 more online plus 3 short diagnostic tests for assessing strengths and areas for improvement and detailed answer explanations for all questions Strengthen your knowledge with in depth review covering all units on the AP Chemistry exam Reinforce your learning with more than 300 practice questions throughout the book that cover all frequently tested topics Learn what to expect on test day with essential details about the exam format scoring calculator policy strategies for all question types and advice for developing a study plan Robust Online Practice Continue your practice with 3 full length practice tests on Barron's Online Learning Hub Simulate the exam experience with a timed test option Deepen your understanding with detailed answer explanations and expert advice Gain confidence with scoring to check your learning progress Power up your study sessions with Barron's AP Chemistry on Kahoot additional free practice to help you ace your exam Learning about Materials Ted Lister, Colin Osborne, 1998 This book presents useful material in the form of teacher's notes and photocopiable worksheets It will be useful at pre and post 16 level **Nature** Sir Norman Lockyer, 1910 Water Chemistry Patrick Brezonik, William Arnold, 2011-03-22 It emphasizes that both equilibrium and kinetic processes are important in aquatic systems **Soil Physical Chemistry** Donald L. Sparks, 2018-02-06 Soil Physical Chemistry Second Edition takes up where the last edition left off With comprehensive and contemporary discussions on equilibrium and kinetic aspects of major soil chemical process and reactions this excellent text reference presents new chapters on precipitation dissolution modeling of adsorption reactions at the mineral water interface and the chemistry of humic substances An emphasis is placed on understanding soil chemical reactions from a microscopic point of view and

rigorous theoretical developments such as the use of modern in situ surface chemical probes such as x ray adsorption fine structure XAFS Fourier transform infrared FTIR spectroscopies and scanning probe microscopies SPM are discussed

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