

# PERIODIC TABLE

## Atomic Properties of the Elements

**NIST**  
National Institute of  
Standards and Technology  
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of <sup>133</sup> Cs																									
speed of light in vacuum c = 299 792 458 m s <sup>-1</sup> (exact)																									
Planck constant h = 6.626 070 15 × 10 <sup>-34</sup> J s (exact)																									
elementary charge e = 1.602 176 634 × 10 <sup>-19</sup> C																									
electron mass m <sub>e</sub> = 9.109 383 56 × 10 <sup>-31</sup> kg																									
proton mass m <sub>p</sub> = 1.672 621 9 × 10 <sup>-27</sup> kg																									
fine-structure constant α = 1/137.036																									
Rydberg constant R <sub>∞</sub> = 10 973 731.568 160 m <sup>-1</sup>																									
Bohrmann constant k <sub>B</sub> = 1.380 658 × 10 <sup>-23</sup> J K <sup>-1</sup>																									

Atomic Number: 58  
Ground-state Ionization Energy (eV): 5.5337

Symbol: **Ce**  
Name: Cerium  
Atomic Weight: 140.116

Ground-state Configuration: [Xe]4f<sup>1</sup>5d<sup>1</sup>6s<sup>2</sup>  
Ionization Energy (eV): 5.5337

\*Based upon <sup>12</sup>C. ( ) indicates the mass number of the longest-lived isotope.

For a description of the data, visit [physics.nist.gov/data](http://physics.nist.gov/data)

NIST SP 966 (September 2010)

# Periodix Table Study Packet

**N Noddings**



## Periodix Table Study Packet:

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the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

**Safety Professional's Reference and Study Guide, Third Edition** W. David Yates, 2020-03-19 This new edition serves both as a reference guide for the experienced professional and as a preparation source for those desiring certifications It s an invaluable resource and a must have addition to every safety professional s library Safety Professional s Reference and Study Guide Third Edition is written to serve as a useful reference tool for the experienced practicing safety professional as well as a study guide for university students and those preparing for the Certified Safety Professional examination It addresses major topics of the safety and health profession and includes the latest version of the Board of Certified Safety Professional BCSP reference sheet a directory of resources and associations as well as state and federal agency contact information Additionally this new edition offers new chapters and resources that will delight every reader This book aids the prospective examination candidate and the practicing safety professional by showing them step by step how to solve each question formula listed on the BCSP examination and provide examples on how and when to utilize them

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