

# GRAPHING TRENDS OF THE PERIODIC TABLE

## CHEMISTRY

NAME: \_\_\_\_\_

DATE: \_\_\_\_\_

PERIOD: \_\_\_\_\_

### Graphing Periodic Table Trends

**Directions:** Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

| Atomic Radius (pm) |     |                 |     |                |     |                |     |
|--------------------|-----|-----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3        | 152 | Sodium - 11     | 190 | Potassium - 19 | 243 | Rubidium - 37  | 265 |
| Beryllium - 4      | 112 | Magnesium - 12  | 145 | Calcium - 20   | 194 | Strontium - 38 | 219 |
| Boron - 5          | 87  | Aluminum - 13   | 118 | Gallium - 31   | 136 | Indium - 49    | 156 |
| Carbon - 6         | 67  | Silicon - 14    | 111 | Germanium - 52 | 125 | Tin - 50       | 140 |
| Nitrogen - 7       | 56  | Phosphorus - 15 | 98  | Arsenic - 33   | 114 | Antimony - 51  | 133 |
| Oxygen - 8         | 48  | Sulfur - 16     | 88  | Selenium - 34  | 103 | Tellurium - 52 | 123 |
| Fluorine - 9       | 42  | Chlorine - 17   | 79  | Bromine - 35   | 94  | Iodine - 53    | 115 |

| Ionization Energy (kJ/mol) |     |                |     |                |     |                |     |
|----------------------------|-----|----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3                | 520 | Sodium - 11    | 496 | Potassium - 19 | 419 | Rubidium - 37  | 403 |
| Beryllium - 4              | 900 | Magnesium - 12 | 738 | Calcium - 20   | 590 | Strontium - 38 | 550 |

### Atomic Radius



# Periodic Trends Graphing Assignment

**Jin-Ying Zhang**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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