#### Periodic Table Assignment

Read pages 14 - 19. Complete the crossword pacete (in pencil) and fill in the period table as instructed below.

ACROSS

DECIMANS

Application in the last of the

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#### Eiverthetic elements with atomic Elements that most olosely numbers greater than 93 followe the premiodic lower This law shales that the properties of. The first extential to organize: elements repeat as a function of alamants based on Missier advantate marridges resonanced terms are off moreons. Billioon as an this period The modern name for etca-Elements were originally placed in **BURBLE DAY** order of atomic \_\_\_\_, now they are in Mont electrorists are perchase of a hourston. cince they full to the left of The group that consists of very Over "submir classes forwer" unreactive garden. 11. The general name given to a 10. Number of elements known in 1800 row (lieft-to-right) in the 13. Electront 101 was named in his periodio teble Programme of the 12: fodine belongs to this group 14. The first allead metal 17. The periodic table is normally drawn with 18 columns. A more sources 15. This element was assigned requirementation has - continuences a refative atomic mass of 19. Most representate are in this state. one because it is the lightest. 20. Nonmetal that is a liquid at SATP Administration . 21. The general name given to a 16. The state of mester column in the periodic table. responses entitled by decah 22. According to figure 6, when 18. Sillicon is in this group. aluminum combines with oxygen. 20. The statement that repainted three oxygen atoms combine with metalis and nomination this many aluminum atoms begins below this element Marie San 29 100 30 80 Section. And Street, mark the same mail to a 100 Profession. mark them BOOK TOWN the beau SHOW SHOW 46 144 66

 Place a large G in boxes that house elements that are gases at NATP, and a large L in boxes that contain elements that are liquids. The rest of the elements are solub; the not write anything in these boxes.

100 100

and to

2007/2004

81 70

100

See Section 1

Acres 1

2) Durken the "staircase line" that divides metals from non-metals

100

The Parket

- Metalloids (i.e. semi-conductors) are elements that have properties midway between metals and non-metals. The metalloids are: B. St. Ce, As. Sb. Tr. Po. At. Identify the metalloids by drawing diagonal lines in these boxes.
- Notice that each group (column) in the periodic table is identified at the top with a number and/or letter. There are two conventions used in the textbook. Label the groups that end in A (e.g. IA, IIA, IIIA, etc).
- Evame and colour the following groups (sens: II is not an IA metal): Alkali metals, alkaline earth metals, halogens, noble gasses, transition metals, the inner transition elements (made up of harbunides and actinides).
- 6) Create a legend for your periodic table.

# **Periodic Table Assignment Answer Key**

**L Darling-Hammond** 

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\*\*College Chemistry I Norman H. Nachtrieb, 1960\*\*

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