



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Answers With Work

Pamela Mills,Amina El-Ashmawy



Percent Yield Answers With Work:

Enjoying the Track of Phrase: An Emotional Symphony within **Percent Yield Answers With Work**

In some sort of taken by displays and the ceaseless chatter of instant communication, the melodic splendor and psychological symphony developed by the prepared word frequently diminish in to the backdrop, eclipsed by the relentless sound and disturbances that permeate our lives. But, situated within the pages of **Percent Yield Answers With Work** a charming literary treasure overflowing with raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that interesting masterpiece conducts visitors on an emotional journey, well unraveling the concealed songs and profound impact resonating within each carefully constructed phrase. Within the depths of the touching examination, we will explore the book is central harmonies, analyze its enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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