

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-3)

Remember: Limiting reagent is the reactant that will completely react and leave another reactant

Excess: All reactants except the limiting reagent will have some left over. Excess is the amount of reactant that is not completely consumed.

Remember: Limiting reagent will determine how much product is produced and by how much. The limiting reagent is also a reactant.



The ratio required for the composition for the limiting reagent.

2H ₂ (g)	20.00 grams
O ₂ (g)	16.00 grams
2H ₂ O(l)	36.00 grams
H ₂ O	18.00 grams

1. 20.00 g of each element will be used & react to form 36.00 g of product water. 18.00 g of water is the limiting reagent. 18.00 g of water is the product.



2. 20.00 g H₂(g) is the limiting reagent. 36.00 g H₂O(l) is the product.

3. 18.00 g of water is the limiting reagent. 18.00 g of water is the product.



4. 18.00 g of water is the limiting reagent. 18.00 g of water is the product.



Percent Yield And Limiting Reactants Quiz

Rachel Sandford



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