

Chemical Formula

Talking about proportions, molecules and relative of atoms present in a compound.

Molecular $\text{C}_{10}\text{H}_{12}\text{O}_4$ (molecular)

Formula represents the number of atoms of each element in a single molecule.

Empirical $\text{C}_5\text{H}_6\text{O}_2$ (molecular)

Formula represents the simplest ratio of positive and negative ions in a compound.

Atomic Mass

The average mass of all atoms of an element is called its atomic mass.

Atomic

1 amu is the mass of a carbon-12 atom.

1 amu is the mass of a proton or a neutron.

Relative Atomic Mass

1 amu is the mass of a carbon-12 atom.

1 amu is the mass of a proton or a neutron.

1 amu is the mass of a proton or a neutron.

Note Taking Guide Episode 601 Chemical Formula Answers

SB Merriam

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape with a gradient, partially overlapping the bar's end.

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