Model: A familiar idea used to explain unfamiliar facts observed in nature.

Theory: An explanation of observable facts and phenomena. To

remain valid, models and theories must:

- explain all known facts
- enable scientists to make correct predictions.

Democritus:

- · proposed the existence of an atom
- word comes from the greek word atomis which means not to cut or "indivisible"

<u>Aristotle</u>

- rejected the idea of the atom
- · said matter could be cut continually

Dalton's theory proposed that atoms:

- are the building blocks of matter
- · are indivisible
- of the same elements are identical
- · of different elements are different
- · unite in small, whole number ratios to form compounds.

J.J. Thomson

- credited with discovery of electron: a blow to Dalton's indivisible atom.
- proposed the plum pudding model of the atom:

Negatively charged electrons embedded in a ball of positive charge

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William Swan Sonnenschein

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Industrial & Engineering Chemistry ,1923 The Journal of Industrial and Engineering Chemistry ,1923 The Chemical Bulletin ,1916 The Chicago Chemical Bulletin ,1914 Books of 1912- ,1922 A Reader's Guide to Contemporary Literature William Swan Sonnenschein,1896 Books of 1921-1925 Chicago Public Library,1927

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