

OTCAS PERSONAL STATEMENT EXAMPLE

I completely believe that all the people deserve nothing but the best quality of life that is possible. They should be allowed to complete their tasks everyday with full ease and satisfaction. It is quite easy to take all of these tasks for granted, not taking into consideration the impact of incapability to complete them. I am personally interested in becoming an occupational therapist because I believe that by adapting environment and teaching others, we can have a positive impact on the life of another person.

I am interested in taking advantage of the opportunity to potentially make a difference by assisting people towards gaining independence, and thus an increase in their self-confidence. I just want to become a part of a very diverse and rewarding career, while having the chance to closely work with other people, assisting them towards reaching progress.

I believe that the current subjects I am currently studying is giving me a good foundation for a degree level study, needing me to be highly logical, while working to meet deadlines. Religious studies and history have taught me in order to look beyond what is obvious, and focusing on underlying aspects.

My social care and health courses have also given me the chance to finish two placements. Both of them have effectively taught me to communicate well with other people of different backgrounds, encouraging me to be open in supporting people in different tasks that can help them to become independent. All of these skills will also help me in forming a better rapport with clients as an occupational therapist.

All in all, I feel that my studies, hobbies, as well as work experience have already provided me with all the vital skills needed in order to become a successful occupational therapist. My caring and hard working nature will also help me to support and improve my service to others.



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G Psacharopoulos



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completed her A Levels in Kolej Tuanku Ja afar where she was awarded the Overall Academic Excellence Book Prize within her cohort She is now a writer and is passionate about travelling stories and finding ways to give back to the community

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s 2 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each

period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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