11-1 Practice Problems

 Lead will react with Hydrochloric acid to produce Lead (II) Chloride and Hydrogen gas. How many moles of Hydrochloric acid are needed to completely react with 0.36-mol of lead? 6. Carbon will react with Zinc Oxide to produce Zinc and Carbon Dioxide. How many moles of carbon dioxide will be produced if 0.38-mol of ZnO is completely reacted?

 How many moles of HNO₃ will be produced when 0.51-mol of N₂O₃ reacts according to the following equation? N₂O₅ + H₂O → 2HNO₃ How many moles of oxygen will be needed to react with 0.38-mol of C₃H₈ according to the following equation?
C₃H₈ + 5O₂ → 3CO₂ + 4H₂O

3. Iron will react with Oxygen Gas to produce Fe₂O₃. How many moles of Fe₂O₃ will be produced if 0.18-mol of Fe reacts? Nitrogen gas can react with Hydrogen gas to produce ammonia. How many moles of Nitrogen will be needed to produce 0.48-mol of NH₃?

 How many moles of NaBr will be produced when 0.69-mol of Bromine reacts according to the following equation?
 Br₂ + 2NaI → 2NaBr + I₂ How many moles of hydrogen will be produced if 0.44-mol of CaH₂ reacts according to the following equation? CaH₂ + 2H₂O → Ca(OH)₂ + 2H₂

5. Phosphorus will react with Bromine to produce Phosphorus Tribromide. How many moles of Phosphorus Tribromide will be produced if 0.78-mol of Bromine is reacted? How many moles of water will be produced if 2.35-mol of oxygen reacts according to the following equation?
 2C₆H₆ + 15O₂ → 12CO₂ + 6H₂O

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