Study Guide

Key Concepts

(14.1 Properties of Gases

- Gases are easily compressed because of the space between particles in a gas.
- The amount of gas (n), volume (V), and temperature (T) are factors that affect gas pressure (P).

C 14.2 The Gas Laws

- · As the pressure of a gas increases, the volume decreases. If the temperature is constant.
- · As the temperature of an enclosed gas increases, the volume increases, if the pressure is constant.
- · As the temperature of an enclosed gas increases, the pressure increases, if the volume is constant.

 The combined gas law allows you to do calculations for situations in which only the amount of gas is constant.

C 14.3 Ideal Gases

- · To calculate the number of moles of a contained gas requires an expression that contains the variable 12.
- · Real gases differ most from an ideal gas at low temperatures and high pressures.

14.4 Gases: Mixtures and Movements

- . In a mixture of gases, the total pressure is the sum of the partial pressures of the gases.
- Gases of lower molar mass diffuse and effuse faster than gases of higher molar mass.

Vocabulary

- Boyle's law (p. 418).
- Charles's law (p. 420)
- combined gas law (p. 424).
- compressibility (p. 413)

- Dalton's law of partial pressures (p. 432)
- diffusion (p. 435).
- effusion (p. 435)
- Gay-Lussac's law (p. 422).

- Graham's law of effusion. (p.435)
- ideal gas constant (p. 426)
- ideal gas law (p. 426)
- partial pressure (p. 432).

Key Equations

- Boyle's law: P₁ × V₁ = P₂ × V₂
- Charles's law: V = V2
- Gay-Lussac's law: P₁ = P₂ / T.
- Combined gas law: $\frac{P_1 \times V_1}{T_1} = \frac{P_2 \times V_2}{T_2}$

- Ideal gas law: P × V = n × R × T or PV = nRT
- Dalton's law: P_{need} = P₁ + P₂ + P₃ + ...
- Graham's law: $\frac{Rate_A}{Rate_n} = \sqrt{\frac{molar \ mass_n}{molar \ mass_A}}$

Organizing Information

Use these terms to construct a concept. map that organizes the major ideas of this chapter.



Concept Map 14 Solve the Concept Map with the help of an interactive guided tutorial.

with Chemiasap

Boyle's Law

Charles's law.

partial pressure

combined gas law.

diffusion

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and learn together i e extending the implementation and knowledge of co design methods In this first edition of our Research Topic on Active Learning we highlight two of the three types of publications we wish to promote First are studies aimed at understanding the pedagogical designs developed by practitioners in their own practices by bringing to bear the theoretical lenses developed and tested in the education research community These types of studies constitute the practice pull that we see as a necessary counterbalance to knowledge push in a more productive pedagogical innovation ecosystem based on research practitioner partnerships Second are studies empirically examining the implementations of evidence based designs in naturalistic settings and under naturalistic conditions Interestingly the teams conducting these studies are already exemplars of partnerships between researchers and practitioners who are uniquely positioned as in between straddling the two worlds As a result these publications represent both the rigours of research and the pragmatism of reflective practice In forthcoming editions we will add to this collection a third type of publication design profiles These will present practitioner developed pedagogical designs at varying levels of abstraction to be held to scrutiny amongst practitioners instructional designers and researchers alike We hope by bringing these types of studies together in an open access format that we may contribute to the development of new forms of practitioner researcher interactions that promote co design in pedagogical Medical Application of Polymer-Based Composites Haw-Ming Huang, 2021-04-21 Polymer based compounds innovation play an important role in modern medical applications Among them high molecular weight polymers modified by biomolecules can increase biological activity and improve their biocompatibility The composite material formed by the biopolymers combined with other materials can improve the mechanical strength These materials that can complement polymers include micron sub micron and nano scale materials Their application covers the entire field of biomedicine The Handbook of Groundwater Engineering Jacques W. Delleur, 2006-11-16 A complete treatment of the theory and practice of groundwater engineering The Handbook of Groundwater Engineering Second Edition provides a current and detailed review of how to model the flow of water and the transport of contaminants both in the unsaturated and saturated zones covers the production of groundwater and the remediation of contaminated groundwater Cultural Transformations in Germania Secunda Berber S. van der Meulen-van der Veen, 2025-07-03 This book uses a multi variate approach to study the Late Roman Lower German frontier challenging the role of migration in settlement changes It highlights long term interactive processes and continuity in material culture styles supported by metallurgical analysis of copper alloy dress ornaments

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