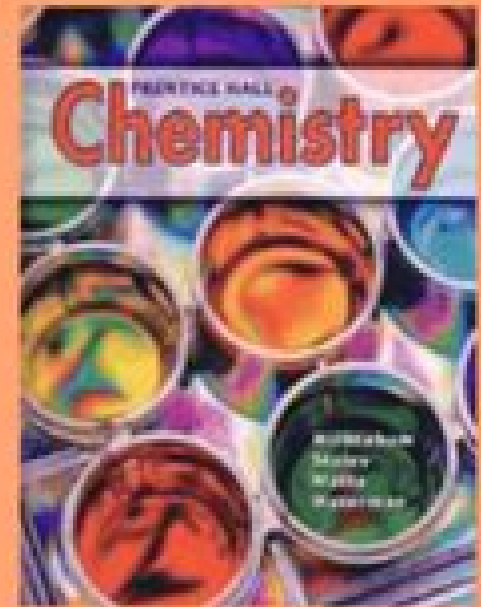


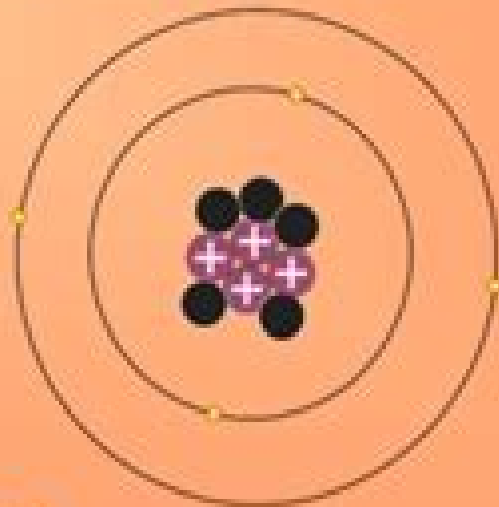
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Chapter 4



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difficult topics such as the wave mechanical atom symmetry elements and symmetry operations and the ligand group orbital approach to bonding The book contains many useful diagrams illustrating among other things the angular dependence of atomic wave functions the derivation of energy level diagrams for polyatomic molecules close packed lattices and ionic crystal structures The diagrams of the periodic variation of atomic and molecular properties showing trends across periods and down groups simultaneously are especially instructive Spectroscopy is presented mainly as a tool for the elucidation of atomic and molecular structures Each chapter begins with a clear and concise statement of What Every First year Student Should Know About outlining the background knowledge that the student is assumed to have from previous courses and thus pointing out what topics might need to be reviewed There are also detailed statements of the objectives of each chapter a number of worked examples interspersed in the text and a comprehensive set of problems and exercises to test the student s understanding Tables of data throughout the text and appendices at the end provide much valuable information National Library of Medicine Current Catalog National Library of Medicine (U.S.), Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office,1974 **El-Hi Textbooks & Serials in Print, 2003** ,2003 **Introduction to Physical Chemistry** Marcus Frederick Charles Ladd,1998-01-22 The third edition of this text has been completely rewritten and revised It is intended for first and second year undergraduates in chemistry taking physical chemistry courses and for undergraduates in other science and engineering subjects that require an understanding of chemistry The author gives more attention to the solid and liquid states than is found in other texts on this subject and introduces topics such as computer simulation and quasicrystals Each chapter concludes with a set of problems to which there are solution notes designed to lead the reader to familiarity with the subject and its application in new situations Computer programs designed to assist the reader are downloadable from the World Wide Web from the time of publication Detailed solutions to the problems will also be available via the World Wide Web See <http://www.cup.cam.ac.uk/stm/ladd/solutions.htm> This modern text on physical chemistry will be of interest to undergraduate students in chemistry and also students in other areas of science and engineering requiring a familiarity with the subject **Environmental Chemistry** Stanley E Manahan,2017-02-24 With clear explanations real world examples and updated questions and answers the tenth edition of Environmental Chemistry emphasizes the concepts essential to the practice of environmental science technology and chemistry while introducing the newest innovations in the field The author follows the general format and organization popular in preceding editions including an approach based upon the five environmental spheres and the relationship of environmental chemistry to the key concepts of sustainability industrial ecology and green chemistry This readily adaptable text has been revamped to emphasize important topics such as the world water crisis It details global climate change to a greater degree than previous editions underlining the importance of abundant renewable energy in minimizing human influences on climate Environmental Chemistry is designed for a wide range of graduate and undergraduate courses in environmental chemistry environmental

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