Table 17.3 Solubilities of lonic Compounds* aq = aqueous (dissolves in water); s = solid (does not dissolve in water)

	Acetate	Bromide	Carbonate	Chlorate	Chloride	Fluoride	Hydrogen Carbonate	Hydroxide	lodide	Nitrate	Nitrite	Phosphate	Sulfate	Sulfide	Sulfite
lons	A	00	0	0	0	E	HO	I	2	Z	Z	-	S	S	ഗ
Aluminum	s	aq		aq	aq	s		s		aq		s	aq		
Ammonium	aq	aq	aq	aq	aq	aq	aq		aq	aq	aq	aq	aq	aq	aq
Barium	aq	aq	S	aq	aq	S		aq	aq	aq	aq	S	s	-	S
Calcium	aq	aq	s	aq	aq	S	0;	s	aq	aq	aq	s	s		s
Cobalt(II)	aq	aq	S	aq	aq	_		S	aq	aq		S	aq	S	S
Copper(II)	aq	aq	S	aq	aq	aq		S		aq		S	aq	S	
Iron(II)	aq	aq	S		aq	S		S	aq	aq		S	aq	S	s
Iron(III)	_	aq			aq	s		s	aq	aq		S	aq	-	
Lead(II)	aq	S	S	aq	S	s		S.	:S	aq	aq	S	S	S	S
Lithium	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	s	aq	aq	aq
Magnesium	aq	aq	s	aq	aq	S		5	aq	aq	aq	S.	aq	16	aq
Nickel	aq	aq	S	aq	aq	aq	1	S	aq	aq		S	aq	S	S
Potassium	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq
Silver	s	s	s	aq	s	aq			s	aq	s	s	s	s	s
Sodium	aq	aq	aq	pg	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq	aq
Zinc	aq	aq	s	aq	aq	aq	10	S	aq	aq		S	aq	S	s

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