

Answer following



- 1. Substrate and temperature 20°C $100, 200, 300$
- 2. A graph of product formed: $100, 200, 300$ (y-axis) vs $100, 200, 300$ (x-axis)
- 3. A graph of substrate and product: $100, 200, 300$ (y-axis) vs $100, 200, 300$ (x-axis)

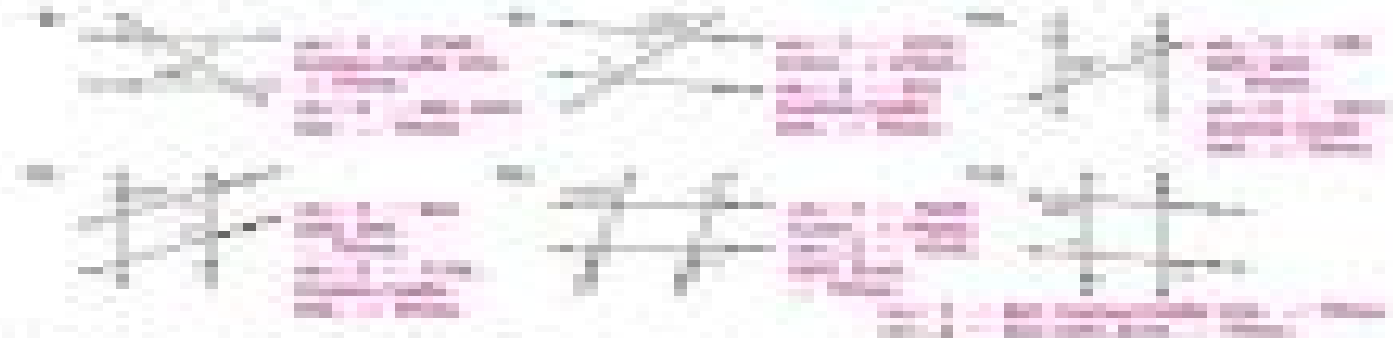
Identify all graphs of rate & time of reaction for the following reaction:
 Reaction: $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$

- 1. concentration of H_2O_2 vs time (y-axis vs x-axis)
- 2. concentration of O_2 vs time (y-axis vs x-axis)
- 3. concentration of H_2O vs time (y-axis vs x-axis)
- 4. concentration of H_2O_2 vs time (y-axis vs x-axis)



Answer: 1, 2, 3, 4

Graphs of rate vs time for the following reaction: $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$



Graphs of rate vs time for the following reaction: $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$



Graphs of rate vs time for the following reaction: $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$

Prentice Hall Geometry 4 5 Answers

C. A. Trapp, Peter Atkins, Julio dePaula

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

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