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Name \_\_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

17

#### THERMOCHEMISTRY

#### SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)

This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.

#### ▶ Energy Transformations (page 505)

- L. What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? <a href="https://doi.org/10.1007/j.com/instry-10.1007/j.com/inst
- Where the use of energy is concerned (in a scientific sense), when is work done?Work is done when a force is used to move an object.
- 3. Circle the letter next to each sentence that is true about energy.
- (a.) Energy is the capacity for doing work or supplying heat.
- (b.) Energy is detected only because of its effects.
- c. Heat is energy that transfers from one object to another because they are at the same temperature.
- (d) Gasoline contains a significant amount of chemical potential energy.
- 4. Circle the letter next to each sentence that is true about heat.
  - a. One effect of adding heat to a substance is an increase in the temperature of that substance.
  - Heat always flows from a cooler object to a warmer object.
  - (c) If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

#### Exothermic and Endothermic Processes (pages 506–507)

What can be considered the "system" and what are the "surroundings" when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: The mixture of chemicals itself is considered the system.

Surroundings: Everything but the mixture of chemicals is the surroundings, but

practically speaking, the immediate vicinity of the system.

# **Prentice Hall Chemistry Section 2 Answers**

Yijin Wang

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Prentice Hall Chemistry .2000 **Computational Chemistry** Errol G. Lewars, 2010-11-10 This corrected second edition contains new material which includes solvent effects the treatment of singlet diradicals and the fundamentals of computational chemistry Computational Chemistry Introduction to the Theory and Applications of Molecular and Quantum Mechanics is an invaluable tool for teaching and researchers alike The book provides an overview of the field explains the basic underlying theory at a meaningful level that is not beyond beginners and it gives numerous comparisons of different methods with one another and with experiment The following concepts are illustrated and their possibilities and limitations are given potential energy surfaces simple and extended Hueckel methods ab initio AM1 and related semiempirical methods density functional theory DFT Topics are placed in a historical context adding interest to them and removing much of their apparently arbitrary aspect The large number of references to all significant topics mentioned should make this book useful not only to undergraduates but also to graduate students and academic and industrial researchers Prentice Hall Science Biology Kenneth Raymond Miller, Prentice Hall (School Division), 1999-02 Explorer Michael J. Padilla, 2002 Prentice Hall Exploring Life Science Anthea Maton, 1997 **Catalog of Copyright Entries. Third Series** Library of Congress. Prentice Hall Science Explorer: Teacher's ed ,2005 Copyright Office, 1978 The Publishers Weekly .1923

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