

## A-LEVEL CHEMISTRY

## MARK SCHEME

## Practice Paper 2

Maximum marks: 101

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Q1.	(a)	(i)	<p><b>M1</b> (Compounds / molecules with) the <u>same structural formula</u>  <i>Penalise M1 if 'same structure' or 'different structural / displayed formula'.</i></p> <p><b>M2</b> with atoms / bonds / groups arranged differently in space  <b>OR</b> atoms / bonds / groups with different spatial arrangements / different orientation  <i>Ignore references to 'same molecular formula' or 'same empirical formula'.</i>  <i>Mark independently.</i></p>	2
		(ii)	 <p><i>Credit C-H<sub>3</sub>C</i>  <i>Credit C<sub>2</sub>H<sub>4</sub></i>  <i>Penalise C-CH<sub>3</sub>CH<sub>3</sub></i></p>	1
	(b)		<p><b>M1</b> Br<sub>2</sub> OR bromine (water) OR bromine (in CCl<sub>4</sub> / organic solvent)  <i>If M1, has no reagent or an incorrect reagent, CE=0.</i>  <i>Ignore 'acidified'.</i></p> <p><b>M2</b> Isomer 1: decolourised / goes colourless / loses its colour  <i>For M1 penalise Br (or incorrect formula of other correct reagent), but mark on.</i></p> <p><b>M3</b> Isomer 2: remains orange / red / yellow / brown / the same OR no reaction / no (observable) change OR reference to colour going to the cyclopentane layer  <i>For M1, it must be a whole reagent and / or correct formula.</i>  <i>If oxidation state given in name, it must be correct. If 'manganate' OR 'manganate(IV)' or incorrect formula, penalise M1, but mark on.</i></p>	1 1 1
			<p><b>Alternatives : potassium manganate(VII)</b></p> <p><b>M1</b> KMnO<sub>4</sub> in acid <b>M2</b> colourless <b>M3</b> purple</p> <p><b>M1</b> KMnO<sub>4</sub> in alkali / neutral <b>M2</b> brown solid <b>M3</b> purple</p> <p><b>Credit for the use of iodine</b></p> <p><b>M1</b> iodine (solution / in KI) <b>M2</b> colourless <b>M3</b> (brown) to purple (credit no change)</p> <p><b>Credit for the use of <u>concentrated</u> H<sub>2</sub>SO<sub>4</sub></b></p>	

# Predicted Paper 2 November 2014 Mark Scheme

**S Marginson**



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