Applied Petroleum Reservoir Engineering 3rd Edition Solution Manual

CHAPTER 2

2.1 using Equation 2.4: pV = nRT

(a)
$$V = \frac{1(10.73)(60 + 460)}{14.7} = 379.6 \text{ SCF}$$

(b)
$$V = \frac{1(10.73)(32+460)}{14.7} = 359.1SCF$$

(c)
$$V = \frac{1(10.73)(80 + 460)}{14.7 + 10/16} = 378.1 \text{ SCF}$$

(d)
$$V = \frac{1(10.73)(60+460)}{15.025} = 371.4 \text{ SCF}$$

- 2.2 (a) number of moles of methane =10 J6 = .625 moles number of moles of ethane = 20/30 = .667 moles total number of moles = .625+.667 = 1.292 moles
- (b) from Equation 2.4

$$p = \frac{1.292(10.73)(550)}{500} = 15.25 \text{ psia}$$

Reservoir Engineering Handbook Solution Manual

J Dewey

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Table of Contents Reservoir Engineering Handbook Solution Manual

- 1. Understanding the eBook Reservoir Engineering Handbook Solution Manual
 - The Rise of Digital Reading Reservoir Engineering Handbook Solution Manual
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Reservoir Engineering Handbook Solution Manual
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reservoir Engineering Handbook Solution Manual
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Reservoir Engineering Handbook Solution Manual
 - Personalized Recommendations
 - Reservoir Engineering Handbook Solution Manual User Reviews and Ratings

- Reservoir Engineering Handbook Solution Manual and Bestseller Lists
- 5. Accessing Reservoir Engineering Handbook Solution Manual Free and Paid eBooks
 - Reservoir Engineering Handbook Solution Manual Public Domain eBooks
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 - Reservoir Engineering Handbook Solution Manual Compatibility with Devices
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 - Highlighting and Note-Taking Reservoir Engineering Handbook Solution Manual
 - Interactive Elements Reservoir Engineering Handbook Solution Manual
- 8. Staying Engaged with Reservoir Engineering Handbook Solution Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reservoir Engineering Handbook Solution Manual
- 9. Balancing eBooks and Physical Books Reservoir Engineering Handbook Solution Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reservoir Engineering Handbook Solution Manual
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reservoir Engineering Handbook Solution Manual
 - Setting Reading Goals Reservoir Engineering Handbook Solution Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reservoir Engineering Handbook Solution Manual
 - Fact-Checking eBook Content of Reservoir Engineering Handbook Solution Manual
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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