

Vicinal dibromide formation:

2. In solvent like H.O.



Halohydrin formation

- + It is a regionelective reaction.
- The nucleophilic attack of the water occurs at the more substituted carbon atom.

2. With Alkymes

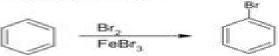
- · It is an anti-addition reaction.
- 1. In solvent like CCI, CH,CI, etc.



Tetrabormide formation

OH

3. With Aromatic Compounds



4. Photo halogenation of alkanes

5. Alpha Bromination of carbonyl compounds

6. Oxidation of Monosaccharides

- Bromine water is a mild oxidizing agent.
- selectively exidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.

HO

OH

OH

7. Haloform Reaction

 converts methyl ketones to Haloform (CHX₃(X=Cl,Br,I)) and salt of carboxylic acids.

8. The Hoffmann Bromamide degradation

 It converts primary amide to a primary amine with one less carbon atom.

9. Hell-Volhard-Zelinski (or HVZ) reaction

5. CN

Cyanide Ion and Nitrile Group

1. Conversion of Nitriles to a Primary Amine

2. Conversion of Nitriles to Aldehydes

3. Reaction of Nitriles with Grignard reagent

4. Hydrolysis of Nitriles to Carboxylic Acids

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Ochem Reagent Guide

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