

**Science (086)**  
**Class X**  
**Sample Question Paper 2022-23**

**Max. Marks: 80**

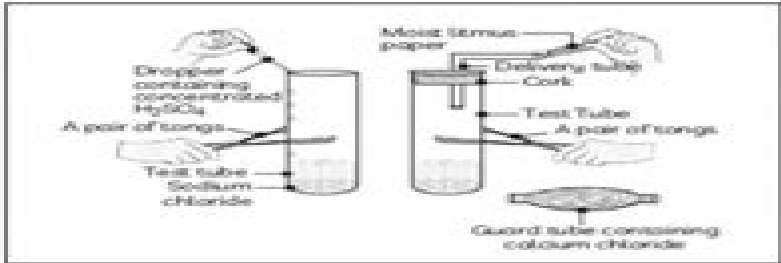
**Time Allowed: 3 hours**

**General Instructions:**

- i. This question paper consists of 39 questions in 5 sections.
- ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. **Section A** consists of 20 objective type questions carrying 1 mark each.
- iv. **Section B** consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- v. **Section C** consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- vi. **Section D** consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. **Section E** consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

**SECTION - A**

Select and write one most appropriate option out of the four options given for each of the questions 1 – 20

| Q. No | Questions                                                                                                                                                                                                                                                                                                                                                                                                     | Marks |
|-------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------|
| 1     | <p>The change in colour of the moist litmus paper in the given set up is due to</p>  <p>i. presence of acid<br/>ii. presence of base<br/>iii. presence of <math>H^+(aq)</math> in the solution<br/>iv. presence of Litmus which acts as an indicator<br/>(a) i and ii<br/>(b) Only ii<br/>(c) Only iii<br/>(d) Only iv.</p> | 1     |
| 2     | <p>In the redox reaction<br/> <math>MnO_2 + 4HCl \rightarrow MnCl_2 + 2H_2O + Cl_2</math><br/>           (a) <math>MnO_2</math> is reduced to <math>MnCl_2</math> &amp; <math>HCl</math> is oxidized to <math>H_2O</math><br/>           (b) <math>MnO_2</math> is reduced to <math>MnCl_2</math> &amp; <math>HCl</math> is oxidized to <math>Cl_2</math></p>                                                 | 1     |

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