

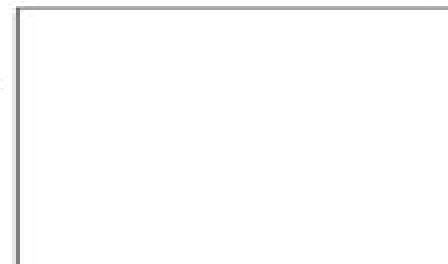
Experiment 1 Report Sheet

Basic Laboratory Operations

Date 03.09.23 Lab Sec. _____ Name XXXXXXXXXX Desk No. _____

A. Bunsen Burner

1. ☒ Instructor's Approval of a well-adjusted Bunsen flame: _____
2.
 - a. At right, draw a sketch of the heat zones for a nonluminous flame as directed with the wire gauze positioned parallel to the burner barrel. Label the "cool" and "hot" zones.
 - b. What happens to the "cool" and "hot" zones of a nonluminous flame, when the air-control valve is closed (nearly closed)? Explain.



B. Laboratory Balances

1. ☒ Determine the mass of the following objects on your assigned balance. Express your results with the correct sensitivity.

Balance no. 11 Sensitivity ± 0.0001

Object	Description of Object (size or volume)	Mass (g)
Test tube	<u>18 × 100</u>	<u>12.9980g</u>
Beaker	<u>100 mL</u>	<u>65.4925g</u>
Spatula	<u>n/a</u>	<u>19.6767g</u>
Graduated cylinder	<u>25 mL</u>	<u>65.4723g</u>
Other (see instructor)	<u>n/a</u>	

2. ☒ Precision of Measurement. Balance no. _____

Mass of graduated cylinder (if not tared) _____

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5	Trial 6
a. Mass of cylinder + H ₂ O (l) (g)	<u>71.0705g</u>	<u>71.074g</u>	<u>71.0700g</u>	<u>71.0743g</u>	<u>71.0788g</u>	<u>71.0730g</u>
b. Mass of 7 mL H ₂ O (l) (g)	<u>6.897g</u>	<u>6.8815g</u>	<u>6.8778g</u>	<u>6.874g</u>	<u>6.8763g</u>	<u>6.8698g</u>
c. Average mass of 7 mL H ₂ O (l) (g)			<u>6.8795g</u>		Data Analysis, B	

Comment on the precision of the mass measurements for the water samples.
There was a lack of consistency for mass of H₂O

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Clifford Lane



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