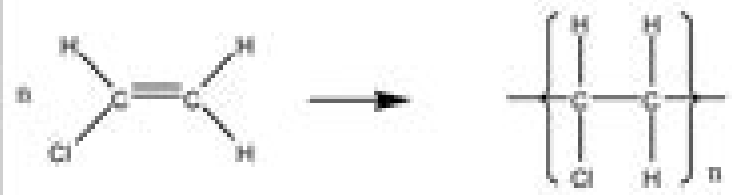


Question	Answer	Mark	Guidance
5 (a)	 <p data-bbox="202 357 637 385">Correct polymer with side links ✓</p> <p data-bbox="202 421 879 492">Balanced equation for formation of correct polymer - correct use of n in the equation and brackets ✓</p>	2	<p data-bbox="1101 135 1883 178">Displayed formulae MUST be used to award each mark</p> <p data-bbox="1101 414 1758 521">n on LHS can be at any height to the left of formula AND n on the RHS must be a subscript (essentially below the side link)</p>
(b)	(i) $\text{CH}_2\text{CHCl} + 2\text{O}_2 \rightarrow \text{CO} + \text{CO}_2 + \text{HCl} + \text{H}_2\text{O}$ ✓	1	<p data-bbox="1101 564 1883 628">ALLOW any other correctly balanced equation with the same reactants and products</p> <p data-bbox="1101 635 1487 664">ALLOW $\text{C}_2\text{H}_2\text{Cl}$ for CH_2CHCl</p>
	(ii) Sodium hydrogencarbonate neutralises HCl ✓	1	<p data-bbox="1101 664 1883 735">Assume that 'it' refers to sodium hydrogencarbonate but DO NOT ALLOW other chemicals e.g. sodium</p> <p data-bbox="1101 771 1883 935">ALLOW NaHCO_3 is a base ALLOW forms a salt or sodium chloride or NaCl ALLOW equation to show formation of NaCl from NaHCO_3 and HCl even if not balanced. IGNORE reacts</p>

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Michael Seilmaier



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