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alkali - ANSWER-A type of base that dissolves in water forming hydroxide ions, OH⁻ (aq) ions.

alkanes - ANSWER-The homologous series with the general formula: C_nH_{2n+2}.

alkyl group - ANSWER-An alkane with a hydrogen atom removed, e.g. CH₃, C₂H₅; alkyl groups are often shown as 'R'.

amount of substance - ANSWER-The quantity whose unit of the mole. Chemists use 'amount of substance' as a means of counting atoms.

anhydrous - ANSWER-A substance that contains no water molecules.

anion - ANSWER-A negatively charged ion.

atom economy - ANSWER-atom economy = $\frac{\text{molecular mass of the desired product}}{\text{sum of molecular masses of all products}} \times 100$

atomic orbital - ANSWER-A region within an atom that can hold up to two electrons, with opposite spins.

atomic (proton) number - ANSWER-The number of protons in the nucleus of an atom

acid - ANSWER-A species that is a proton donor.

activation energy - ANSWER-The minimum amount of kinetic energy that particles need to have in order to react when they collide

addition polymer - ANSWER-A very long molecular chain formed by repeated addition reactions of many unsaturated alkene molecules (monomers).

addition polymerisation - ANSWER-The process in which unsaturated alkene molecules (monomers) add on to a growing polymer chain one at a time to form a very long saturated molecular chain (the addition polymer).

addition reaction - ANSWER-A reaction in which a reactant is added to an unsaturated molecule to make a saturated molecule.

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none



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