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(Chapter - 1) (Matter in our Surroundings) (Exercises) (Class - IX)

Question 1:

Convert the following temperatures to the Celsius scale.

(a) 293 K

(b) 470 K

Answer 1:

- (a) $293 \text{ K} = (293 273)^{\circ}\text{C} = 20 ^{\circ}\text{C}$
- (b) $470 \text{ K} = (470 273)^{\circ}\text{C} = 197 ^{\circ}\text{C}$

Ouestion 2:

Convert the following temperatures into the Kelvin scale.

(a) 25°C

(b) 373 °C

- _Answer 2:
- (a) 25 °C = (25 + 273) K = 298 K
- (b) $373 \,^{\circ}\text{C} = (373 + 273) \,^{\circ}\text{K} = 646 \,^{\circ}\text{K}$

Ouestion 3:

Give reasons for the following observations.

- (a) Naphthalene balls disappear with time without leaving any solid.
- (b) We can get the smell of perfume sitting several meters away.
- _Answer 3:

(a) Naphthalene shows the property of sublimation. Evaporation of naphthalene takes place easily and so it disappears during course of time without leaving a solid.

(b) Perfumes vaporize very fast and its vapours diffuse into air easily. That is why we can smell perfume sitting several meters away.

Ouestion 4:

Arrange the following substances in increasing order of forces of attraction between the particles water, sugar, oxygen.

Answer 4:

Oxygen < Water < Sugar.

Ouestion 5:

What is the physical state of water at -

(a) 25°C

(b) 0°C

(c) 100°C?

- Answer 5:
- (a) Liquid
- (b) Solid and Liquid
- (c) Liquid and Vapours

Question 6:

Give two reasons to justify:

- (a) Water at room temperature is a liquid.
- (b) An iron almirah is a solid at room temperature.

Answer 6:

(a) Water at room temperature is a liquid because it has fluidity and has definite volume but no definite

(b) An iron almirah is a solid at room temperature because it is rigid and has a definite shape.

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Table of Contents Ncert Science Class Solutions Chapter 1

- 1. Understanding the eBook Ncert Science Class Solutions Chapter 1
 - The Rise of Digital Reading Ncert Science Class Solutions Chapter 1
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Ncert Science Class Solutions Chapter 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ncert Science Class Solutions Chapter 1
 - o User-Friendly Interface
- 4. Exploring eBook Recommendations from Ncert Science Class Solutions Chapter 1
 - Personalized Recommendations
 - Ncert Science Class Solutions Chapter 1 User Reviews and Ratings
 - Ncert Science Class Solutions Chapter 1 and Bestseller Lists
- 5. Accessing Ncert Science Class Solutions Chapter 1 Free and Paid eBooks
 - Ncert Science Class Solutions Chapter 1 Public Domain eBooks
 - Ncert Science Class Solutions Chapter 1 eBook Subscription Services
 - Ncert Science Class Solutions Chapter 1 Budget-Friendly Options
- 6. Navigating Ncert Science Class Solutions Chapter 1 eBook Formats

- o ePub, PDF, MOBI, and More
- Ncert Science Class Solutions Chapter 1 Compatibility with Devices
- Ncert Science Class Solutions Chapter 1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ncert Science Class Solutions Chapter 1
 - Highlighting and Note-Taking Ncert Science Class Solutions Chapter 1
 - Interactive Elements Ncert Science Class Solutions Chapter 1
- 8. Staying Engaged with Ncert Science Class Solutions Chapter 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ncert Science Class Solutions Chapter 1
- 9. Balancing eBooks and Physical Books Ncert Science Class Solutions Chapter 1
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Ncert Science Class Solutions Chapter 1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ncert Science Class Solutions Chapter 1
 - Setting Reading Goals Ncert Science Class Solutions Chapter 1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ncert Science Class Solutions Chapter 1
 - Fact-Checking eBook Content of Ncert Science Class Solutions Chapter 1
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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