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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this
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declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95
82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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condition at which a phase change occurs from liquid to vapour or vapour to liquid

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the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates
the volume and the number of particles in each flask indicates relative amounts