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$$3 \text{ KClO}_3 \rightarrow 3 \text{ KCl} + 3 \text{ O}_2$$

$$2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2$$

$$4 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$$

$$5 \text{ Pb} + 2 \text{ OH}^- + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$$

$$2 \text{ Al} + 3 \text{ Br}_2 \rightarrow 2 \text{ AlBr}_3$$

$$3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 \rightarrow 3 \text{ H}_2\text{O} + 2 \text{ CO}_2$$

$$4 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$$

$$8 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$$

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