

Ideal Gases

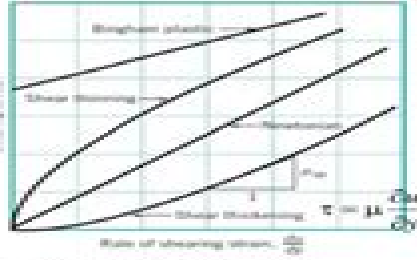
- From chemistry: $PV = nRT$ (V is volume)
 - $n = m / M$ is the number of moles
 - for mass in kg n is in kilogram moles (kmol); for mass in lb_m, n is in pound moles (lbmol)
 - $R = 8.31447$ kJ/kmol-K = 10.7316 psia-ft³ / lbmol-R is universal gas constant
 - $R = R/M$ is engineering gas constant that is different for each gas
 - Real gases like ideal gases at low pressures
 - $P = nRT / V = (m/M)RT / V = (m/V)(R/M)T$
 $P = \rho RT$



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Viscosity

Newtonian and non-Newtonian
Variation of shear stress with rate of shearing strain for several types of fluids, including common non-Newtonian fluids.





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Surface Tension Rise



- Vertical force balance: $\gamma \pi R^2 h = 2 \pi R \sigma \cos \theta$
 - Surface tension depends on fluid and temperature, wetting angle, θ , depends on fluid and surface

$$h = \frac{2 \sigma \cos \theta}{\gamma R}$$



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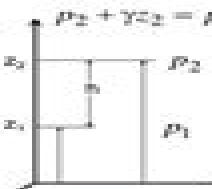
Pressure Relations

- Pressure is a scalar
- The force exerted by a pressure is the same in all directions
- Want to see how pressure changes in a static (nonmoving fluid)
- Look at balance of pressure force and fluid weight over a differential volume element, $\delta x \delta y \delta z$



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Incompressible Fluid



- Which pressure is higher?
 $p_1 = p_2 + \gamma(z_2 - z_1)$
 $p_1 = p_2 + \gamma h > p_2$
- Pressure increases with depth



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Gage and Absolute Pressure

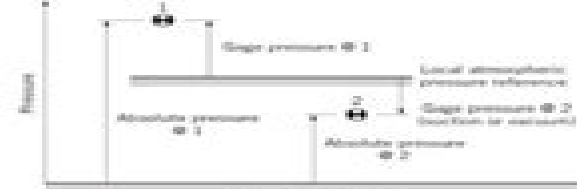


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Review For Final Exam Fluid Mechanics

Dilip K. Das, Rajaram K. Prabhudesai



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Monthly Catalog of United States Government Publications ,1983 Annual Catalogue United States Air Force Academy,1985 Professional Engineer ,1985 A Brief Introduction to Fluid Mechanics Donald F. Young,Bruce R. Munson,Theodore H. Okiishi,Wade W. Huebsch,2010-11-23 A Brief Introduction to Fluid Mechanics 5th Edition is designed to cover the standard topics in a basic fluid mechanics course in a streamlined manner that meets the learning needs of today s student better than the dense encyclopedic manner of traditional texts This approach helps students connect the math and theory to the physical world and practical applications and apply these connections to solving problems The text lucidly presents basic analysis techniques and addresses practical concerns and applications such as pipe flow open channel flow flow measurement and drag and lift It offers a strong visual approach with photos illustrations and videos included in the text examples and homework problems to emphasize the practical application of fluid mechanics principles *Chemical Engineering* Dilip K. Das,Rajaram K. Prabhudesai,2004 This is a review book for people planning to take the PE exam in Chemical Engineering Prepared specifically for the exam used in all 50 states It features 188 new PE problems with detailed step by step solutions The book covers all topics on the exam and includes easy to use tables charts and formulas It is an ideal desk companion to DAS s Chemical Engineer License Review It includes sixteen chapters and a short PE sample exam as well as complete references and an index Chapters include the following topical areas Material and energy balances Fluid dynamics Heat transfer Evaporation Distillation Absorption Leaching Liq liq extraction Psychrometry and humidification

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