

DESCRIBING CHEMICAL REACTIONS

Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation

Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical 1.

The substances that undergo a chemical change are the 2. The new substances formed in a chemical reaction are the 3. In accordance with the law of conservation of 4, a chemical equation must be balanced.

When balancing an equation, you place 5 in front of reactants and products so that the same number of atoms of each 6 are on each side of the equation. An equation must never be balanced by changing the 7 in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is 8; for a solid, 9; and for a gas, 10. A substance dissolved in water is designated 11. If a 12 is used to increase the rate of a chemical reaction, its formula is written above the arrow.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____
11. _____
12. _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- _____ 14. The symbol Δ placed over the arrow in an equation means that heat is supplied to the reaction.
- _____ 15. Atoms are destroyed in a chemical reaction.
- _____ 16. A skeleton equation is not a balanced equation.

Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I

- a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.
- b. Acetylene gas (C_2H_2) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

Review And Reinforce Describing Chemical Reactions Answers

LL Leslie

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Table of Contents Review And Reinforce Describing Chemical Reactions Answers

1. Understanding the eBook Review And Reinforce Describing Chemical Reactions Answers
 - The Rise of Digital Reading Review And Reinforce Describing Chemical Reactions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Review And Reinforce Describing Chemical Reactions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Review And Reinforce Describing Chemical Reactions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Review And Reinforce Describing Chemical Reactions Answers
 - Personalized Recommendations
 - Review And Reinforce Describing Chemical Reactions Answers User Reviews and Ratings
 - Review And Reinforce Describing Chemical Reactions Answers and Bestseller Lists

5. Accessing Review And Reinforce Describing Chemical Reactions Answers Free and Paid eBooks
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 - Review And Reinforce Describing Chemical Reactions Answers eBook Subscription Services
 - Review And Reinforce Describing Chemical Reactions Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Review And Reinforce Describing Chemical Reactions Answers Compatibility with Devices
 - Review And Reinforce Describing Chemical Reactions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review And Reinforce Describing Chemical Reactions Answers
 - Highlighting and Note-Taking Review And Reinforce Describing Chemical Reactions Answers
 - Interactive Elements Review And Reinforce Describing Chemical Reactions Answers
8. Staying Engaged with Review And Reinforce Describing Chemical Reactions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review And Reinforce Describing Chemical Reactions Answers
9. Balancing eBooks and Physical Books Review And Reinforce Describing Chemical Reactions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review And Reinforce Describing Chemical Reactions Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Review And Reinforce Describing Chemical Reactions Answers
 - Setting Reading Goals Review And Reinforce Describing Chemical Reactions Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Review And Reinforce Describing Chemical Reactions Answers
 - Fact-Checking eBook Content of Review And Reinforce Describing Chemical Reactions Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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