REVERSIBLE REACTIONS

- Every reaction can proceed in two directions: forward and reverse
- Mercury (II) oxide decomposes when heated
 - $2HgO(s) \rightarrow 2Hg(l) + O_2(g)$
- Mercury and oxygen combine to form mercury (II) oxide when heated gently
 - $2Hg(I) + O_2(g) \rightarrow 2HgO(s)$

Reversible Reactions 18 Answers

WJ Hussar

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