Atomic Theory & Structure: Atomic Theory & Isotope Skill Review

- 1) State the five parts to Dalton's Atomic Theory.
 - 1. All elements are made of atoms
 - 2. All atoms of an element are identical
 - 3. All atoms of different elements are different
 - 4. I aw of constant composition
 - 5. Law of conservation of matter
- Describe I.J. Thomson's theory of atomic structure including the experiment he performed to reach this theory.

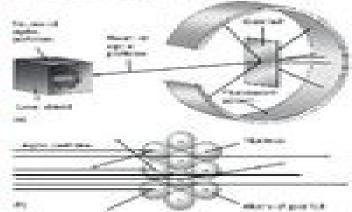
After conducting his research using a gasfilled cathode ray tube where Thomson pumped electricity through the gas sample, he was able to conclude that atoms release electrons/contain regatively charged particles called electrons.



Atoms are a cloud of positive charges with electrons randomly distributed throughout.

Describe Ennest Rutherford's theory of atomic structure including the experiment he performed to reach this theory. Include a picture.

Butherford conducted the gold-fold experiment to confirm the Thomson's theory of atomic structure. The experiment was dione with a radioactive source of alpha marticles aimed at a piece of gold foil with a detection screen to locate positively charged alpha particles either when they pass through the gold atoms or are deflected. The result's showed that most of the alpha particles were detected behind the foil and only a small percentage deflected. After calculating deflection patterns, the only aloba particles that deflected were those that came close to the center of the atom. Based on the results. But berford made three conclusions: 13 Plum pudding model is incorrect. 20 the atom has a positively



charged center, 30 the rest of the atom (where electrons are located) is empty space.

Review And Reinforcement Atomic Theory Answer

Michal Rosen-Zvi

Review And Reinforcement Atomic Theory Answer:

Unveiling the Magic of Words: A Review of "Review And Reinforcement Atomic Theory Answer"

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