

SUBJECT: SCIENCE
CLASS : X

MAX. MARKS: 80
DURATION: 3 HRS

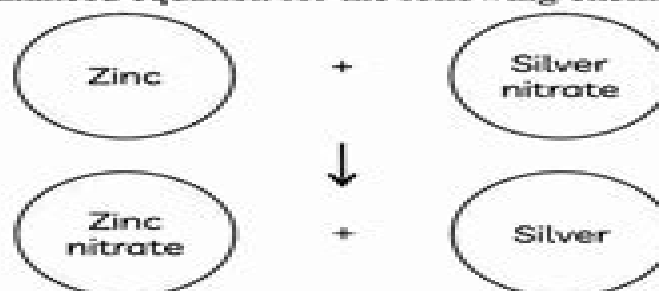
General Instruction:

1. This Question Paper has 5 Sections A-E.
2. **Section A** has 20 MCQs carrying 1 mark each.
3. **Section B** has 5 questions carrying 02 marks each.
4. **Section C** has 6 questions carrying 03 marks each.
5. **Section D** has 4 questions carrying 05 marks each.
6. **Section E** has 3 case based integrated units of assessment (04 marks each) with sub-parts of the values of 1, 1 and 2 marks each respectively.
7. All Questions are compulsory. However, an internal choice in 2 Qs of 5 marks, 2 Qs of 3 marks and 2 Questions of 2 marks has been provided. An internal choice has been provided in the 2marks questions of Section E
8. Draw neat figures wherever required. Take $\pi = 22/7$ wherever required if not stated.

SECTION – A

Questions 1 to 20 carry 1 mark each.

1. Which of the following a balanced equation for the following chemical reaction:



- (a) $\text{Zn} + \text{Ag}(\text{NO}_3)_2 \rightarrow \text{Zn}(\text{NO}_3)_2 + 2\text{Ag}$
 (b) $2\text{Zn} + 2\text{AgNO}_3 \rightarrow \text{Zn}(\text{NO}_3)_2 + 2\text{Ag}$
 (c) $\text{Zn} + 2\text{AgNO}_3 \rightarrow \text{Zn}(\text{NO}_3)_2 + 2\text{Ag}$
 (d) $\text{Zn} + \text{AgNO}_3 \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Ag}$
2. Which of the following will turn phenolphthalein pink?
 (a) $\text{NaOH}(\text{aq})$ (b) $\text{HCl}(\text{aq})$ (c) $\text{CH}_3\text{COOH}(\text{aq})$ (d) H_2O
3. While studying the saponification reaction, what do you observe when you mix an equal amount of colourless vegetable oil and 20% aqueous solution of NaOH in a beaker?
 (a) The colour of the mixture has become dark brown.
 (b) A brisk effervescence is taking place in the beaker.
 (c) The outer surface of the beaker has become hot.
 (d) The outer surface of the beaker has become cold.
4. Why do we store silver chloride in dark-coloured bottles?
 (a) To prevent precipitation of silver chloride
 (b) To prevent decomposition of silver chloride
 (c) To promote decomposition of silver chloride
 (d) All of these

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Table of Contents Sample Papers Of Evt

1. Understanding the eBook Sample Papers Of Evt
 - The Rise of Digital Reading Sample Papers Of Evt
 - Advantages of eBooks Over Traditional Books
2. Identifying Sample Papers Of Evt
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sample Papers Of Evt
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Papers Of Evt
 - Personalized Recommendations
 - Sample Papers Of Evt User Reviews and Ratings
 - Sample Papers Of Evt and Bestseller Lists
5. Accessing Sample Papers Of Evt Free and Paid eBooks
 - Sample Papers Of Evt Public Domain eBooks
 - Sample Papers Of Evt eBook Subscription Services
 - Sample Papers Of Evt Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Sample Papers Of Evt Compatibility with Devices
 - Sample Papers Of Evt Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sample Papers Of Evt
 - Highlighting and Note-Taking Sample Papers Of Evt
 - Interactive Elements Sample Papers Of Evt
8. Staying Engaged with Sample Papers Of Evt
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sample Papers Of Evt
9. Balancing eBooks and Physical Books Sample Papers Of Evt
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sample Papers Of Evt
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sample Papers Of Evt
 - Setting Reading Goals Sample Papers Of Evt
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sample Papers Of Evt
 - Fact-Checking eBook Content of Sample Papers Of Evt
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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