

TUTORIAL 3

ALKENES & CYCLOALKENES

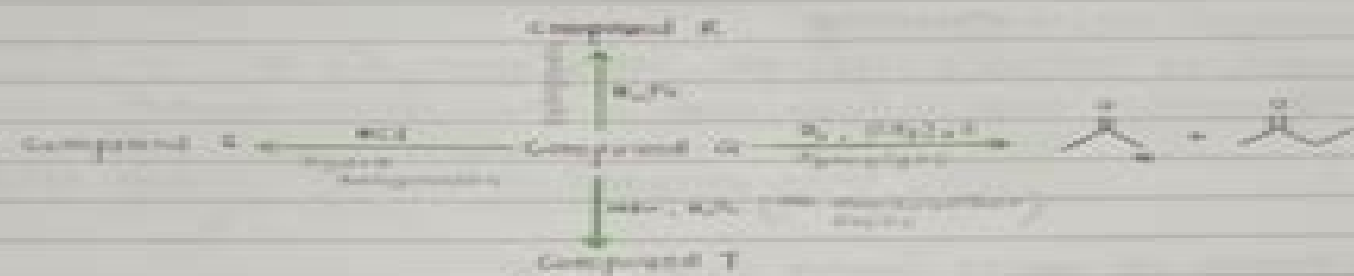
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Q1

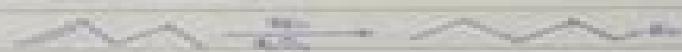
Alkene can undergo various reactions. Identify the structure of R, S, T and U.



Compound R (cis-2-butene)



Compound T (anti-2-butanol diol)



Compound S (2-methylbut-2-ene)



Compound U (2-methylbut-1-ene)



Q2

a) Define hydrocarbon compounds.

A class of organic compounds made up of only the elements carbon (C) and hydrogen (H).

b) Indicate the difference between propene and propane

Propene	Propane
- Propene is a saturated hydrocarbon compound.	- Propane is a unsaturated hydrocarbon compound.
- carbon-carbon double bond.	- carbon-carbon single bond in the structure.
- fully saturated to bond.	- double bond and single bond.

Reaction Lab Answer Key Hydrocarbon

Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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