

Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

18.998

Na

Sodium

24.305

Mg

Magnesium

39.098

K

Potassium

40.078

Ca

Calcium

44.956

Sc

Scandium

47.88

Ti

Titanium

50.942

V

Vanadium

51.996

Cr

Chromium

54.938

Mn

Manganese

55.935

Fe

Iron

58.933

Co

Cobalt

58.933

Ni

Nickel

63.546

Cu

Copper

65.38

Zn

Zinc

69.723

Ga

Gallium

72.64

Ge

Germanium

74.922

As

Arsenic

78.971

Se

Selenium

79.904

Br

Bromine

83.80

Kr

Krypton

85.468

Rb

Rubidium

87.62

Sr

Strontium

88.906

Y

Yttrium

91.224

Zr

Zirconium

92.906

Nb

Niobium

95.94

Mo

Molybdenum

97.905

Tc

Technetium

101.07

Ru

Ruthenium

102.906

Rh

Rhodium

106.42

Pd

Palladium

107.868

Ag

Silver

112.411

Cd

Cadmium

114.818

In

Indium

118.710

Sn

Tin

121.757

Sb

Antimony

127.76

Te

Tellurium

126.905

I

Iodine

131.29

Xe

Xenon

132.905

Cs

Cesium

137.327

Ba

Barium

138.905

La

Lanthanum

175.053

Hf

Hafnium

180.948

Ta

Tantalum

183.84

W

Tungsten

186.207

Re

Rhenium

188.906

Os

Osmium

190.23

Ir

Iridium

195.084

Pt

Platinum

196.967

Au

Gold

200.592

Hg

Mercury

204.379

Tl

Thallium

207.2

Pb

Lead

208.980

Bi

Bismuth

209

Po

Polonium

210

At

Astatine

210

Rn

Radon

223

Fr

Francium

226

Ra

Radium

227

Ac

Actinium

261

Rf

Rutherfordium

262

Ha

Hassium

263

Sg

Seaborgium

263

Bh

Berkelium

264

Hs

Hassium

265

Mt

Mendelevium

289

(113)

289

(115)

289

(117)

1

1.01

H

Hydrogen

3

6.94

Li

Lithium

4

9.01

Be

Beryllium

11

22.99

Na

Sodium

12

24.31

Mg

Magnesium

19

39.10

K

Potassium

20

40.08

Ca

Calcium

21

44.96

Sc

Scandium

22

47.88

Ti

Titanium

23

50.94

V

Vanadium

24

51.99

Cr

Chromium

25

54.94

Mn

Manganese

26

55.94

Fe

Iron

27

58.93

Co

Cobalt

28

58.93

Ni

Nickel

29

63.55

Cu

Copper

30

65.39

Zn

Zinc

31

69.72

Ga

Gallium

32

72.64

Ge

Germanium

33

74.92

As

Arsenic

34

78.97

Se

Selenium

35

79.90

Br

Bromine

36

83.80

Kr

Krypton

37

85.47

Rb

Rubidium

38

87.62

Sr

Strontium

39

88.91

Y

Yttrium

40

91.22

Zr

Zirconium

41

92.91

Nb

Niobium

42

95.94

Mo

Molybdenum

43

97.91

Tc

Technetium

44

101.07

Ru

Ruthenium

45

102.91

Rh

Rhodium

46

106.42

Pd

Palladium

47

107.87

Ag

Silver

48

112.41

Cd

Cadmium

49

114.82

In

Indium

50

118.71

Sn

Tin

51

121.76

Sb

Antimony

52

127.76

Te

Tellurium

53

126.91

I

Iodine

54

131.30

Xe

Xenon

55

132.91

Cs

Cesium

56

137.33

Ba

Barium

57

138.91

La

Lanthanum

58

175.05

Hf

Hafnium

59

180.95

Ta

Tantalum

60

183.85

W

Tungsten

61

186.21

Re

Rhenium

62

188.91

Os

Osmium

63

190.24

Ir

Iridium

64

195.08

Pt

Platinum

65

196.97

Au

Gold

66

200.59

Hg

Mercury

67

204.38

Tl

Thallium

68

207.21

Pb

Lead

69

208.98

Bi

Bismuth

70

209.0

Po

Polonium

71

210.0

At

Astatine

72

210.0

Rn

Radon

73

223.0

Fr

Francium

74

226.0

Ra

Radium

75

227.0

Ac

Actinium

76

261.0

Rf

Rutherfordium

77

262.0

Ha

Hassium

78

263.0

Sg

Seaborgium

79

263.0

Bh

Berkelium

80

264.0

Hs

Hassium

81

265.0

Mt

Mendelevium

82

289.0

(113)

83

289.0

(115)

84

289.0

(117)

atomic number

14

28.086

Si

Silicon

name

symbol

black

solid

blue

liquid

red

gas

white

synthetically prepared

most stable isotope

alkali metals

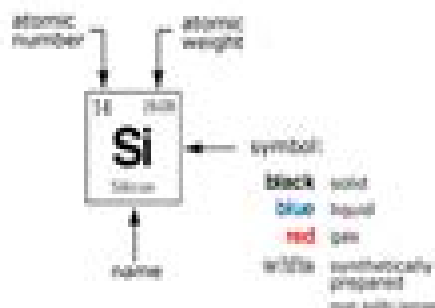
alkaline earth metals

transitional metals

other metals

nonmetals

noble gases



58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

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AnOrthodoxUnderstandingOfTheBibleWithPhysicalScience htm

Hume's Natural Philosophy and Philosophy of Physical Science Matias Slavov,2020-10-15 This book contextualizes David Hume s philosophy of physical science exploring both Hume s background in the history of early modern natural philosophy and its subsequent impact on the scientific tradition Drawing on Cartesian cosmology and Einstein s special relativity and taking in topics including experimentalism

causation laws of nature metaphysics of forces mathematics relation to nature and the concepts of space and time this book deepens our understanding of Hume's relation to natural philosophy It does so in addition by situating Hume's thought within the context of other major philosophers and scientists including Descartes Locke Boyle Kant Newton and Leibniz Demonstrating above all Hume's understanding of the fluid relationship between philosophy and science Hume's *Natural Philosophy* and *Philosophy of Physical Science* will provide new insights for historians and philosophers of science

Chemical news and Journal of physical science, 1866 **The Chemical News and Journal of Physical Science**, 1904 I. *The Greek school philosophy, with reference to physical science.* II. *The physical sciences in ancient Greece.* III. *Greek astronomy.* IV. *Physical science in the middle ages.* V. *Formal astronomy after the stationary period.* VI. *Mechanics, including fluid mechanics.* VII. *Physical astronomy.* Additions to the 3d ed William Whewell, 1858 *A Review of the Progress of Mathematical and Physical Science in More Recent Times, and Particularly between the Years 1775 and 1850* James D. Forbes, 2025-08-27 Reprint of the original first published in 1858 The Antigonos publishing house specialises in the publication of reprints of historical books We make sure that these works are made available to the public in good condition in order to preserve their cultural heritage **Proceedings of the Royal Society. Section A, Mathematical and**

Physical Science Royal Society (Great Britain), 1911 *The Philosophy of Physical Science* Sir Arthur Stanley Eddington, 1939 *Peterson's Grad Programs in Physical Sciences, Math, Ag Sciences, Envir & Natural Res 20154 (Grad 4)* Peterson's, 2014-10-21 Graduate Programs in the Physical Sciences Mathematics Agricultural Sciences the Environment Natural Resources 2015 contains more than 3 000 graduate programs in the relevant disciplines including agriculture and food sciences astronomy and astrophysics chemistry physics mathematics environmental sciences and management natural resources marine sciences and more Informative data profiles for more than 3 000 graduate programs at nearly 600 institutions are included complete with facts and figures on accreditation degree requirements application deadlines and contact information financial support faculty and student body profiles Two page in depth descriptions written by featured institutions offer complete details on specific graduate programs schools or departments as well as information on faculty research Comprehensive directories list programs in this volume as well as others in the graduate series **Chemistry**

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Climate Change 2007 - The Physical Science Basis Intergovernmental Panel on Climate Change, Intergovernmental Panel on Climate Change. Working Group I, 2007-09-10 The Climate Change 2007 volumes of the Fourth Assessment Report of the Intergovernmental Panel on Climate Change IPCC provide the most comprehensive and balanced assessment of climate change available This IPCC Working Group I report brings us completely up to date on the full range of scientific aspects of climate change Written by the world's leading experts the IPCC volumes will again prove to be invaluable for researchers students and policymakers and will form the standard reference works for policy decisions for government and industry worldwide

Cæsar, for beginners [selections from books 1, 4, 5 and 6] with notes by J. Currie Gaius Julius Caesar, 1854

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how to understand the meaning of research terms how to disseminate research in a style which is clear and easily
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