

# Physical Science EOCT Study Guide – Section 3

## Physics: Energy, Force, and Motion

Standards: PFS.3 and 4

- i. Students will analyze transformations and flow of energy within a system.
  - a. Explain thermal energy (heat) and describe the principle of conservation of energy as it applies to the law of entropy.
  - b. Explain heat. Theoretical physics heat flows from cold to hot. Heat is transferred by conduction, convection, and radiation.
  - c. Explain specific heat. The amount of heat needed to raise the temperature of 1 kg of a substance by 1°C.
  - d. What is the formula for determining that change in heat that energy can hold?
 
$$Q = mc\Delta T$$

$$\Delta T = T_2 - T_1$$
  - e. A sample of 2.0 kg of water at 20°C and specific heat of 4186 J/kg°C. Find the change in heat energy if the block is 2 mmol from 10°C to 20°C.
 
$$Q = mc\Delta T$$

$$20 = 2000(4186)(10)$$

$$8372000 \text{ J}$$
  - f. Explain conduction, convection, and radiation. Give examples of each.
 Conduction is the transfer of heat through direct contact. Convection is the transfer of heat through the movement of a fluid. Radiation is the transfer of heat through electromagnetic waves.

- ii. Explain the following phase change diagrams.



- iii. Students will determine relationships among force, mass and motion.
  - a. Explain the difference between distance and displacement.
 Distance is the total length of the path traveled. Displacement is the change in position from the starting point to the ending point.
  - b. What is the difference between average speed and instantaneous speed?
 Average speed is the total distance divided by the total time. Instantaneous speed is the speed at a particular moment in time.
  - c. What is the difference between acceleration and velocity?
 Velocity is the speed in a given direction. Acceleration is the rate of change of velocity.

$$v = \frac{d}{t}$$

# Physical Science Eoct Study Guide Energy Transformations

**JL Elias**



## **Physical Science Eoct Study Guide Energy Transformations:**

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## **Table of Contents Physical Science Eoct Study Guide Energy Transformations**

1. Understanding the eBook Physical Science Eoct Study Guide Energy Transformations
  - The Rise of Digital Reading Physical Science Eoct Study Guide Energy Transformations
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Eoct Study Guide Energy Transformations
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Eoct Study Guide Energy Transformations
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Eoct Study Guide Energy Transformations
  - Personalized Recommendations
  - Physical Science Eoct Study Guide Energy Transformations User Reviews and Ratings
  - Physical Science Eoct Study Guide Energy Transformations and Bestseller Lists
5. Accessing Physical Science Eoct Study Guide Energy Transformations Free and Paid eBooks
  - Physical Science Eoct Study Guide Energy Transformations Public Domain eBooks
  - Physical Science Eoct Study Guide Energy Transformations eBook Subscription Services
  - Physical Science Eoct Study Guide Energy Transformations Budget-Friendly Options
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  - ePub, PDF, MOBI, and More
  - Physical Science Eoct Study Guide Energy Transformations Compatibility with Devices
  - Physical Science Eoct Study Guide Energy Transformations Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Eoct Study Guide Energy Transformations
  - Highlighting and Note-Taking Physical Science Eoct Study Guide Energy Transformations
  - Interactive Elements Physical Science Eoct Study Guide Energy Transformations
8. Staying Engaged with Physical Science Eoct Study Guide Energy Transformations

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Physical Science Eoct Study Guide Energy Transformations
- 9. Balancing eBooks and Physical Books Physical Science Eoct Study Guide Energy Transformations
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Eoct Study Guide Energy Transformations
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Eoct Study Guide Energy Transformations
  - Setting Reading Goals Physical Science Eoct Study Guide Energy Transformations
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Eoct Study Guide Energy Transformations
  - Fact-Checking eBook Content of Physical Science Eoct Study Guide Energy Transformations
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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