

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion always forms with a charge of 2-? _____ (anion)
Which cation is always a transition metal? _____ (cation)
2. An element in the halogen group has _____ valence electrons, and tends to _____ 1 electron to have 8. This is because it has _____ valence.
3. An _____ bond is the attraction between oppositely charged ions. HCl is a chemical bond is _____.
4. An ionic bond is formed between _____, which is attracted by an ionic bond, whereas _____, which is not.
5. Elements in the _____ family tend to be reduced easily by a relatively small amount of _____, and hence they're reduced easily to the _____ state, which is a common state. In fact, they're reduced to a common state by losing only one electron.
6. An ionic bond is formed by _____ valence electrons. This family includes all the periodic table elements that are _____, which is reduced.
7. Elements in the _____ family tend to be reduced by a relatively small amount of _____.

B. Formulas for Ionic Compounds - Fill in the blank below

Write the correct chemical formula for the compound.	Aluminum	Barium	Potassium	Lithium
Chlorine				
Sulfur				

C. Write the chemical formula for each of the following ions:

Write your answer below the Lewis structure, and use arrows to show the transfer of electrons. Use the periodic table to find the oxidation number. They will combine to form a neutral compound. Write the chemical formula for each compound.

Physical Science Chemical Bonding Study Guide

Scott C. Dulebohn



Physical Science Chemical Bonding Study Guide:

Whispering the Secrets of Language: An Psychological Quest through **Physical Science Chemical Bonding Study Guide**

In a digitally-driven world where monitors reign great and quick conversation drowns out the subtleties of language, the profound secrets and mental subtleties concealed within words often move unheard. Yet, set within the pages of **Physical Science Chemical Bonding Study Guide** a captivating literary value sporting with raw thoughts, lies a fantastic quest waiting to be undertaken. Published by a talented wordsmith, that enchanting opus attracts viewers on an introspective trip, softly unraveling the veiled truths and profound impact resonating within the very cloth of each and every word. Within the mental depths of this emotional review, we will embark upon a honest exploration of the book is core themes, dissect its interesting writing type, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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