

Name: Michaela Hubig

Matr.:

Klausur: Physikalische Chemie

Physikalische Chemie: Chapter 10: Outer Thermal Energy and Mass

Multiple Choice:

1. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
2. Energy added to a system:
- a) increases the system's internal energy
 - b) decreases the system's internal energy
 - c) increases the system's temperature
 - d) decreases the system's temperature
3. When a system is heated, the temperature of the system:
- a) increases
 - b) decreases
 - c) remains constant
 - d) increases or decreases depending on the system
4. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
5. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
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- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
7. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
8. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
9. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$
10. The Boltzmann constant is defined as:
- a) $k_B = 1.38 \times 10^{-23} \text{ J/K}$
 - b) $1.38 \times 10^{-23} \text{ J/K}$
 - c) $1.38 \times 10^{-23} \text{ J/K}$
 - d) $1.38 \times 10^{-23} \text{ J/K}$

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