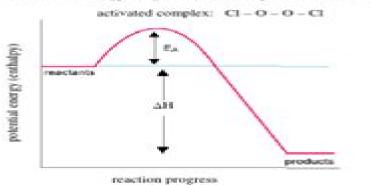
Unit 4, Lesson 03: Collision Theory and the Rates of Chemical Reactions Homework

Page 294, Q 13 - 16

13. Reaction: $2 \text{ ClO }(g) \rightarrow \text{ Cl}_2(g) + \text{ O}_2(g)$

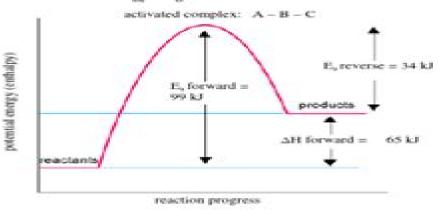
Potential Energy Diagram for Decomposition of CiO (g)



14. Reaction: $AB + C \rightarrow AC + B$ $\Delta H = +65 \text{ kJ}$, $E_a \text{ reverse} = 34 \text{ kJ}$

The potential energy diagram is the correct shape, but it is not drawn accurately to scale.

Potential Energy Diagram for Reaction of AB with C



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Table of Contents Reaction Rates Lesson 36 Answers

- 1. Understanding the eBook Reaction Rates Lesson 36 Answers
 - The Rise of Digital Reading Reaction Rates Lesson 36 Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Reaction Rates Lesson 36 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reaction Rates Lesson 36 Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Reaction Rates Lesson 36 Answers

- Personalized Recommendations
- Reaction Rates Lesson 36 Answers User Reviews and Ratings
- Reaction Rates Lesson 36 Answers and Bestseller Lists
- 5. Accessing Reaction Rates Lesson 36 Answers Free and Paid eBooks
 - Reaction Rates Lesson 36 Answers Public Domain eBooks
 - Reaction Rates Lesson 36 Answers eBook Subscription Services
 - Reaction Rates Lesson 36 Answers Budget-Friendly Options
- 6. Navigating Reaction Rates Lesson 36 Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Reaction Rates Lesson 36 Answers Compatibility with Devices
 - Reaction Rates Lesson 36 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reaction Rates Lesson 36 Answers
 - Highlighting and Note-Taking Reaction Rates Lesson 36 Answers
 - Interactive Elements Reaction Rates Lesson 36 Answers
- 8. Staying Engaged with Reaction Rates Lesson 36 Answers
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reaction Rates Lesson 36 Answers
- 9. Balancing eBooks and Physical Books Reaction Rates Lesson 36 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reaction Rates Lesson 36 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reaction Rates Lesson 36 Answers
 - Setting Reading Goals Reaction Rates Lesson 36 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reaction Rates Lesson 36 Answers

- Fact-Checking eBook Content of Reaction Rates Lesson 36 Answers
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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