

Pre-Lab Information

Purpose	Explore the rate of a chemical reaction at different temperatures and with different reactant particle sizes.
Time	Approximately 45 minutes
Question	What are the effects of temperature and a reactant's particle size on reaction rate?
Hypothesis 1	If you increase the temperature of a reaction, then the reaction rate will increase because particles experience more collisions at higher temperatures.
Hypothesis 2	If you decrease the particle size of a reactant, then the reaction rate will increase because more of the reactants' surface area is exposed allowing more particles to make contact with each other.
Variables	Independent variables: particle size, temperature Dependent variable: reaction rate
Summary	The reaction between effervescent tablets and water will produce CO_2 gas, which you can observe in the form of bubbles; when the bubbling stops, the reaction is complete. You will measure the reaction times and rates for tablets dissolved at four temperatures and broken into three different particle sizes.

Safety

- Always wear a lab coat and safety goggles when performing an experiment.
- Behavior in the lab needs to be purposeful. Use caution when using the hot plate, as it can cause burns.
- Check glassware such as graduated cylinders and beakers for cracks and chips prior to use.
- Use the right gear, such as chemically resistant gloves and oven mitts, when performing the experiment.
- Report all accidents—no matter how big or small—to your teacher.

Procedure

Step 1: Gather Materials

- 250 mL graduated cylinder
- Thermometer
- Water
- Timer
- Four 250 mL beakers
- Seven 1,000 mg effervescent tablets
- Two pieces of filter paper
- 600 mL beaker
- Ice
- Hot plate

Reaction Rates Chemical Study Guide

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Reaction Rates Chemical Study Guide:

The book delves into Reaction Rates Chemical Study Guide. Reaction Rates Chemical Study Guide is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Reaction Rates Chemical Study Guide, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Reaction Rates Chemical Study Guide
- Chapter 2: Essential Elements of Reaction Rates Chemical Study Guide
- Chapter 3: Reaction Rates Chemical Study Guide in Everyday Life
- Chapter 4: Reaction Rates Chemical Study Guide in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Reaction Rates Chemical Study Guide. This chapter will explore what Reaction Rates Chemical Study Guide is, why Reaction Rates Chemical Study Guide is vital, and how to effectively learn about Reaction Rates Chemical Study Guide.
3. In chapter 2, the author will delve into the foundational concepts of Reaction Rates Chemical Study Guide. The second chapter will elucidate the essential principles that need to be understood to grasp Reaction Rates Chemical Study Guide in its entirety.
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5. In chapter 4, the author will scrutinize the relevance of Reaction Rates Chemical Study Guide in specific contexts. The fourth chapter will explore how Reaction Rates Chemical Study Guide is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Reaction Rates Chemical Study Guide. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Reaction Rates Chemical Study Guide.

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