Types of Chemical Reactions

Name____

Period_____Date____

<u>Directions:</u> Label the following reactions as synthesis (S), decomposition (D), single replacement (SR), double replacement (DR) and combustion (C), then balance. Place a "B" at the end that are balanced as written.

3.
$$C_2H_4 + O_2 \rightarrow CO_2 + H_2O$$

4.
$$H_2O \rightarrow H_2 + O_2$$

5.
$$H_2 + O_2 \rightarrow H_2O$$

6.
$$C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$$

11.
$$C_6H_{12} + O_2 \rightarrow CO_2 + H_2O$$

18.
$$C_6H_8 + O_2 \rightarrow CO_2 + H_2O$$

20.
$$H_2 + N_2 \rightarrow NH_3$$

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