

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

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Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

**NATIONAL
SENIOR CERTIFICATE**

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in Reaction IV.



4.8 Draw the half-cell reaction for the zinc electrode.



4.9 Identify the reaction.

Write a balanced chemical equation for the reaction.

4.10 Draw a half-cell reaction for the zinc electrode.

4.11 The simplified diagram below represents an electrochemical cell. The electrodes are made of copper.

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L Reisser



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Table of Contents Physical Science Paper2 2014 November Memorandum

1. Understanding the eBook Physical Science Paper2 2014 November Memorandum
 - The Rise of Digital Reading Physical Science Paper2 2014 November Memorandum
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper2 2014 November Memorandum
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Paper2 2014 November Memorandum
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper2 2014 November Memorandum

- Personalized Recommendations
 - Physical Science Paper2 2014 November Memorandum User Reviews and Ratings
 - Physical Science Paper2 2014 November Memorandum and Bestseller Lists
5. Accessing Physical Science Paper2 2014 November Memorandum Free and Paid eBooks
- Physical Science Paper2 2014 November Memorandum Public Domain eBooks
 - Physical Science Paper2 2014 November Memorandum eBook Subscription Services
 - Physical Science Paper2 2014 November Memorandum Budget-Friendly Options
6. Navigating Physical Science Paper2 2014 November Memorandum eBook Formats
- ePub, PDF, MOBI, and More
 - Physical Science Paper2 2014 November Memorandum Compatibility with Devices
 - Physical Science Paper2 2014 November Memorandum Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Paper2 2014 November Memorandum
 - Highlighting and Note-Taking Physical Science Paper2 2014 November Memorandum
 - Interactive Elements Physical Science Paper2 2014 November Memorandum
8. Staying Engaged with Physical Science Paper2 2014 November Memorandum
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Paper2 2014 November Memorandum
9. Balancing eBooks and Physical Books Physical Science Paper2 2014 November Memorandum
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Paper2 2014 November Memorandum
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper2 2014 November Memorandum
- Setting Reading Goals Physical Science Paper2 2014 November Memorandum
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Paper2 2014 November Memorandum

- Fact-Checking eBook Content of Physical Science Paper2 2014 November Memorandum
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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