

D 14%

1.5 There are ... atoms of hydrogen in 36g of H_2O .

- A 2.4×10^{24}
- B 6×10^{23}
- C 4.8×10^{24}
- D 6×10^{24}

[10]

QUESTION 2

An aeroplane flew in the following manner: 200km west, 150km south and then 180km east.

- 2.1 ~~Draw a vector diagram to show the displacement of the plane.~~ (3)
- 2.2 ~~Calculate the displacement of the plane.~~ (3)
- 2.3 ~~What is the final velocity of the plane if it starts from rest and accelerates at 2 m/s^2 for 10s?~~ (3)
- 2.4 If the plane took 15min to complete this journey, determine the average speed it flew by. (3)

[10]

QUESTION 3

A car drips oil. The driver pulls out of his driveway and leaves a set of oil drops on the road as he travels. The oil drops are dropped every half a second. The distance of the drops can be recorded as follows:

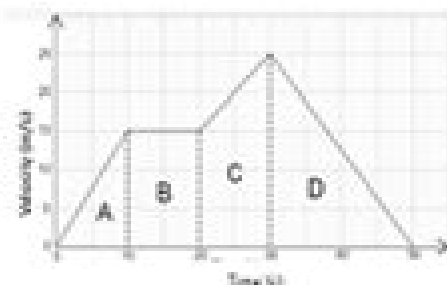
Drop point	Distance from starting point
A	0m
B	1m
C	4m
D	9m
E	16m
F	25m

- 3.1 Use the graph to calculate the acceleration of the car. (5)
- 3.2 ~~Calculate the average velocity of the car between points A and F.~~ (3)
- 3.3 Calculate the magnitude of the velocity at point E. (3)
- 3.4 ~~Calculate the average velocity of the car between points A and F.~~ (4)

[10]

QUESTION 4

The velocity vs. time graph below illustrates a part of a mom's journey, dropping her kids off at school.



NO EQUATIONS OF MOTION SHOULD BE USED IN ANSWERING THE FOLLOWING QUESTIONS.

- 4.1 ~~Calculate the acceleration of the vehicle between 0s and 10s.~~ (3)
- 4.2 ~~What is the speed of the vehicle at 20s?~~ (1)
- 4.2 Use the graph to calculate the acceleration at the following intervals:
 - 4.2.1 ~~Between 0s and 10s.~~ (2)
 - 4.2.2 ~~Between 10s and 20s.~~ (2)
- 4.3 Use the graph to calculate the total displacement of the vehicle. (10)

[10]

QUESTION 5

A vehicle starts from rest and accelerates in a southerly direction at 3 m/s^2 .

- 5.1 ~~Calculate the velocity of the vehicle after 10s.~~ (2)
- 5.2 ~~Calculate the distance travelled by the vehicle in 10s.~~ (3)
- 5.3 ~~Calculate the velocity of the vehicle after 10s.~~ (3)
- 5.4 ~~Calculate the average velocity of the vehicle.~~ (3)

[10]

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web consider the thermochemical equation below $4\text{NO(g)} + 6\text{H}_2\text{O(l)} + 4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 11\text{H}_2\text{O(l)} + 7\text{N}_2\text{(g)}$ is this reaction endothermic or exothermic endothermic $\Delta H^\circ = 0$ b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of H_2

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web heats of fusion and vaporization determines what determines how much energy is required to change a substance's phase what is the heat of fusion formula $q = m \cdot \Delta H_{\text{fus}}$ what is the heat of vaporization formula $q = m \cdot \Delta H_{\text{vap}}$ what is on the x axis of the heating curve time

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web sep 5 2023 3 carry out a thermochemical calculation on all these species at their optimized geometries 4 sum up the enthalpies free energy of the products and the reagents keeping into account the

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore ΔH must also be halved 3 answer c explanation option c is incorrect because 566 kJ/mol is the ΔH per mole of O_2 and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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web thermochemistry practice worksheet answer key what is the heat change when 472 g of carbon reacts with excess O_2 according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat $2\text{C}_3\text{O}_2 + 2\text{S} \rightarrow 3\text{H}_2 + 7914\text{kJ}$ is this endothermic or exothermic 2

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web thermochemistry the study of the transfer of heat thermal energy in chemical reactions exothermic transfer of heat from the system to the surroundings $2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$ energy endothermic the transfer of heat from the surroundings to the system energy $2\text{Hg(s)} + 2\text{Hg(l)} + \text{O}_2\text{(g)}$

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web q 140 g 4 184 j 16oc g oc q 9372 16 j 35 5 g x 1 mole kcn 0 5453 moles 65 1 g 0 5453 mole moles 5 when a 15 g sample of SrCl_2 dissolves in 80 g of water in a calorimeter the temperature drops from 42 oc to 15 c calculate the Δh for the process

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