

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and which from it? (anion/cation) _____
which cation it gains and electron it loses from it? _____ (anion/cation)
2. Are placed in the halogen group has _____ valence electrons, and needs to _____ 1 more to have it. This is because it has _____ valence.
3. Are _____ bond to the adjacent elements oppositely charged ions. (PCl₃ is covalent bond is _____)
4. Is a covalent bond, shared _____ electrons, achieved by the bond bond, all are _____ electrons.
5. (Electron group) Which side from left shared completely by it completely bond is to _____, and which group is shared equally to the _____ bond group it is covalent. In ionic, gain electrons, or negative charged by, losing losing from _____
6. Are ionic ions have up to _____ valence electrons. This ionic bond is the particular bond that all have 8 valence electrons, or called _____
7. Shared up to eight, have _____, or called _____ by the other side of bond.

B. Formulas for Ionic compounds - Put in the right order

Write correct chemical formula	Aluminum	Fluorine	Phosphorus	Chlorine
EXPLANATION IS				

C. Write the chemical formulas for each of the following ions: compound it

Write your own ionic compounds structures, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.

Physical Science Chemical Bonding Study Guide

Léo Pomar



Physical Science Chemical Bonding Study Guide:

Unveiling the Magic of Words: A Report on "**Physical Science Chemical Bonding Study Guide**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Physical Science Chemical Bonding Study Guide**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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