

### Chapter 6: Chemical Bonds Review Answer Key

- What is the most stable group on the Periodic Table? Why is this group stable?  
**Noble Gases (Group 18) because they have 8 valence electrons which means their outer energy level is full.**
- Name the most reactive group(s) on the Periodic Table? Why are they reactive?  
**Alkali Metals (Group 1) & Halogens (Group 17): Alkali Metals want to lose one valence electron in order to become stable and Halogens want to gain one valence electron to be stable.**

- In the boxes below draw the Electron dot diagrams for:

| Sodium      | Calcium                 | Boron                    | Chlorine                                    | Krypton                                     |
|-------------|-------------------------|--------------------------|---------------------------------------------|---------------------------------------------|
| <b>Na ·</b> | <b>Ca ·</b><br><b>·</b> | <b>· B ·</b><br><b>·</b> | <b>· ·</b><br><b>· : Cl ·</b><br><b>· ·</b> | <b>· ·</b><br><b>· : Kr :</b><br><b>· ·</b> |

- Name the three (3) types of chemical bonds:  
**Ionic bonds      Covalent bonds      Polyatomic**
- Define ionic bond:  
**The force that holds cations and anions together.**
- Define covalent bond:  
**A chemical bond in which two atoms share a pair of valence electrons.**
- Define polyatomic ion:  
**A covalently bonded group of atoms that has a positive or negative charge and acts as a unit.**
- In an ionic bond, electrons are transferred.
- Who are the participants in an ionic bond? **Metals and nonmetals**
- An ion with a positive charge is a cation; an ion with a negative charge is an anion.
- In a covalent bond, electrons are shared.
- Who are the participants in a covalent bond? **Two nonmetals**
- Identify whether the following compounds are covalent (two nonmetals), ionic (a nonmetal and a metal), or polyatomic (both).
  - CaCl<sub>2</sub>      IONIC (transferred)**
  - H<sub>2</sub>O      COVALENT (shared)**
  - BaSO<sub>4</sub>      POLYATOMIC (both)**
  - CH<sub>4</sub>      COVALENT (shared)**
  - LiBr      IONIC (transferred)**
  - NH<sub>4</sub>Cl      POLYATOMIC (both)**
  - KI      IONIC (transferred)**

# Physical Science Chapter 6 Chemical Bonds Wordwise Answers

**Guillaume Favre**



## **Physical Science Chapter 6 Chemical Bonds Wordwise Answers:**

## Whispering the Strategies of Language: An Emotional Journey through **Physical Science Chapter 6 Chemical Bonds Wordwise Answers**

In a digitally-driven earth wherever displays reign great and instant communication drowns out the subtleties of language, the profound techniques and mental subtleties hidden within words usually go unheard. However, nestled within the pages of **Physical Science Chapter 6 Chemical Bonds Wordwise Answers** a fascinating fictional prize blinking with organic feelings, lies an extraordinary quest waiting to be undertaken. Published by a talented wordsmith, that wonderful opus invites viewers on an introspective journey, lightly unraveling the veiled truths and profound impact resonating within the fabric of each and every word. Within the emotional depths with this emotional review, we shall embark upon a heartfelt exploration of the book is primary subjects, dissect its charming writing type, and yield to the powerful resonance it evokes strong within the recesses of readers hearts.

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