

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is.
 - a. always positive.
 - b. always negative.
 - c. either positive or negative.
 - d. zero.

- _____ 2. According to the octet rule, a calcium atom has a tendency to.
 - a. lose one electron.
 - b. lose two electrons.
 - c. gain one electron.
 - d. gain two electrons.

- _____ 3. If a compound forms by ionic bonding, which is *not* true?
 - a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
 - b. The net charge of the compound is zero.
 - c. The compound contains just two atoms, each of opposite charge.
 - d. Several ions group together in a tightly packed structure.

- _____ 4. The only property listed that is *not* characteristic of ionic compounds is.
 - a. high melting point.
 - b. hardness.
 - c. lack of crystal structure.
 - d. good conductor of electricity.

- _____ 5. Which formula listed below represents a polyatomic ion?
 - a. NH_4^+
 - b. H_2SO_4
 - c. NaCl
 - d. H_2O

- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because.
 - a. ionic substances tend to vaporize at room temperature.
 - b. ionic substances are brittle.
 - c. attractive forces between ions are greater than the attractive forces between molecules.
 - d. the numbers of positive and negative charges are equal in an ionic compound.

Physical Science Chapter 6 Chemical Bonds Test Review

SJ Ball

Physical Science Chapter 6 Chemical Bonds Test Review:

Whispering the Strategies of Language: An Mental Quest through **Physical Science Chapter 6 Chemical Bonds Test Review**

In a digitally-driven world wherever screens reign supreme and quick conversation drowns out the subtleties of language, the profound strategies and mental nuances hidden within phrases usually get unheard. Yet, located within the pages of **Physical Science Chapter 6 Chemical Bonds Test Review** a charming literary value blinking with organic thoughts, lies an extraordinary quest waiting to be undertaken. Composed by a skilled wordsmith, this enchanting opus attracts visitors on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within ab muscles material of every word. Within the psychological depths with this emotional evaluation, we shall embark upon a genuine exploration of the book is key styles, dissect its interesting writing design, and fail to the effective resonance it evokes serious within the recesses of readers hearts.

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